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REPORT

ON THE

EXPLORATION POTENTIAL

OF THE

MUSCOVITE - GRAPHITE - GOLD DEPOSIT

OF

THE QUINTO MINING CORPORATION,

VERNON MINING DIVISION,

BRITISH COLUMBIA, CANADA

LOCATED

NEAR LUMBY, B.C.

50 DEGREES 15.9 MINUTES NORTH LATITUDE 118 DEGREES 56.3 MINUTES WEST LONGITUDE N.T.S. MAP AREA 82L/7W, 82L/6E

CONSISTING OF 19 CLAIMS (251 units) (B.S. 1-2, 4-5, P.S., P.S. 2-4, P.S. 7, QUIN, M.M. 1-5, LUM 1-4)

FOR

THE QUINTO MINING CORPORATION 606-626 WEST PENDER STREET VANCOUVER, BRITISH COLUMBIA V6B 1V9

BY

A.D. Drummond, Ph.D., P.Eng. GEOLOGICAL BRANCH D.A. Howard, MASS.SE.S.MENT REPORT

March 9, 1993

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### SUMMARY

The Quinto Mining Corporation owns 251 mineral claim units covering Saddle Mountain and the area north of the town of Lumby, B.C. in the Vernon Mining Division of British Columbia. Road access is excellent to and on the property. Local elevation of Saddle Mountain is from 500 metres (1,640 ft. asl.) to about 900 metres (about 3,000 ft. asl.). Geomorphologically the area has rounded hills, sparse outcrop and is covered with pine, cedar, shrubs and mountain grasses. Coordinates are 50 degrees 15.9 minute north latitude and 118 degrees 56.3 minutes west longitude. N.T.S. area is 82L/6E and 7W.

The mineral claim group covers Upper Triassic aged sedimentary and metamorphic rocks and has been intruded by Jurassic and Cretaceous intrusions. The Saddle Mountain portion of The Quinto Mining Corporation property has been the site of most of the The southern part of Saddle Mountain is exploration work. underlain by a hornblende diorite of Cretaceous age which has intruded a Upper Triassic sequence of argillaceous - micaceous schist rocks which near the hornblende diorite intrusion dip at about 45 degrees to the south toward the intrusion. It is within the tilted argillaceous horizon that the Plateau Shear Zone has formed and which is the site of the herein described muscovitegraphite-gold mineralization. Other zones of mineralization occur within the claim group and are mentioned but the emphasis of this report is directed to the evaluation of the potential economic worth of the muscovite, graphite and gold within the Plateau Shear Zone.

The Plateau Shear Zone has been traced across Saddle Mountain in an east-west direction for about 1,000 metres (3,280 ft.) by drilling. The muscovite, graphite and gold mineralization within the shear zone has been confirmed down dip in excess of 150 metres (about 500 ft.) and is still open. The down dip length from the ridge of Saddle Mountain to the valley floor is 500 metres (1,640 ft.).

Gold mineralization occurs as very fine grained native gold within pyrite grains and along hairline fractures in the abundant fine grained to massive pyrite. Recent sampling of the underground workings by the writer shows that the 190 X-cut contains a 20 metre interval (zone wider but not sampled) averaging 0.044 O.P.T. gold with an assay range of 0.001 to 0.313 O.P.T. gold. The 140 X-cut averaged 0.060 O.P.T. over a 14 metre width with an assay range of 0.005 to 0.222 O.P.T. gold. All samples were at least 2 metres in width. A comparison with previous sampling confirmed that an analytical repeatability problem still exists which indicated that only by collecting bulk samples will the problem be addressed.

Previous investigations had suggested that the Plateau Shear Zone contained a very high percentage of graphite because of its

shiny black appearance. After extensive metallurgical testing, microscope study, and research into the analytical methods used to analyze for graphite it was found that the flotation product only contained 5 to 6 percent graphite, therefore it is probable that rock the original contained somewhat less. The graphite (identified with a petrographic microscope - 1000 power) is ultra fine grained (0.1-0.3 by 2-5 microns) and occurs as interleaved grains between fine grained muscovite/sericite grains. The reason that the high percentage of muscovite floated was that the graphite adheres to some of the muscovite thus making it readily floatable. The economic significance of this phenomenon is that it is a very effective way to produce a high grade muscovite product which is valuable in its own right.

Metallurgical testing (Flotation using DF- 250 only) of the graphitic material produced a concentrate that represented 15 percent by weight of the original sample. The concentrate assayed 0.136 O.P.T. gold and contained 5.35 percent graphite with the remainder mainly very fine grained muscovite/sericite. Further flotation (with an addition of a collector) produced a pyrite concentrate which represented 20 percent by weight of the original The tail (62.8 percent by sample and assayed 0.98 O.P.T. gold. sample) assayed 0.038 O.P.T. gold. of weight original The calculated head assay is 0.275 verses an assay head of 0.18 O.P.T. aold. This upgrading factor is consistent with previous metallurgical work.

Further work is recommended to better define (1) the character and composition of the fine grained muscovite product with emphasis on producing a product that equals or exceeds the ASTM standards for fine grained muscovite, (2) the leachability of gold prior to flotation of the muscovite/graphite product, (3) methods of low temperature removal of the ultra-fine graphite from the muscovite product (ultra-fine grained graphite oxidizes at a much lower temperature than coarse flake graphite) and finally (4) a program of bulk testing to define the actual gold grade within the deposit.

### INTRODUCTION

The firm of D.D.H. Geomanagement Ltd., 422-470 Granville Street, Vancouver, B.C., V6C 1V5 has been requested by The Quinto Mining Corporation, 606-626 West Pender Street, Vancouver, B.C., V6B 1V9 to supervise and report on recent investigations on possible economic potential of the newly defined graphite/mica association (micaceous graphitic schist) with known gold mineralization. Prior investigations dealt only with the gold content in the zone.

To accomplish this assignment, all available data both public and private has been reviewed. The property has been visited and sampled by the writer and numerous samples have been submitted for assay and metallurgical testing.

### LOCATION, ACCESS AND INFRASTRUCTURE

The Lumby gold-graphite deposit (Figures 1 and 2) of The Quinto Mining Corporation is located in south central British Columbia just outside the city, but in part within the village limits of Lumby, B.C. and is traversed by B.C. Highway No. 6.

Access to the property is via a network of narrow two and/or four wheel drive old logging roads. A good gravel road is in place to the underground workings and the 150 ton mill.

The town of Lumby is a service centre for local logging and agriculture located 22 kilometres east of Vernon, B.C. The town contains all necessary facilities in terms of lodging, restaurants and fuel to support any exploration program. A 3 phase power line extends from Lumby to the above mill and could be activated upon request.

#### PROPERTY AND TITLE

The Lumby property comprises 251 units contained in 19 modified grid claims (See Figure 3). The claims all lie within the Vernon Mining Division on N.T.S. map sheets 82L/7W and 82L/6E. The claims are listed below:

CLAIM	I N	AME	RECORD NO.	UNITS	MIN. TEN. NO.	EXPIRY DATE
B.S.	-	1	2002	20	259244	9/24/93
B.S.	-	2	2003	20	259245	9/24/93
B.S.	-	4	2005	20	259247	9/24/93
B.S.	-	5	2006	20	259248	9/24/93
P.S.			-	20	308526	4/01/93
P.S.	-	2	2007	20	259249	9/24/93
P.S.	-	3	2008	20	259250	9/24/93
P.S.	_	4	2009	20	259251	9/24/93
P.S.	-	7	2012	15	259254	9/24/93
QUIN			3536	10	260147	5/07/94
м.м.	-	1	_	1	308527	3/31/93
М.М.	-	2	-	1	308528	3/31/93
М.М.	÷ .	3	-	1	308529	3/31/93
М.М.		4	-	1	308530	3/31/93

м.м.	-	5	-	1	308531	3/31/93
LUM	-	1		20	313856	9/26/93
LUM	-	2	-	20	313865	9/28/93
LUM	-	3	-	20	313866	9/28/93
LUM	-	4	-	1	304469	9/27/93
			TOTAL	251		

#### HISTORY

Historical development of mining activity on Saddle Mountain adjacent to Lumby, B.C. is summarized from Kuran (1986), Lebel (1987) and Bradley (1990).

Mineralization at the south end of Saddle Mountain adjacent to Lumby, B.C. was noted in the early 1900's by a local teacher whose prospect workings have been named the Teacher Showing.

After a hiatus of some 50 years, Chaput Logging Company exposed silver-lead-zinc-copper veins during logging operations on the west side of Saddle Mountain in the 1960's at which time the showing was staked (since named the Mine Showing). In 1968, F.K. Explorations Ltd. acquired the claims covering this showing, started underground development and constructed a 50 tonne per day flotation mill. During the period 1968 - 1970, some 1500 tonnes of concentrate were shipped to the smelter at Trail, B.C. Work was terminated.

In 1971, Alberta Gypsum Ltd. acquired the property and mill, undertook underground and surface exploration in an attempt to establish mineable reserves. Work was terminated in 1973. Coast Interior Ventures Ltd. acquired the property in 1974 and operated sporadically in the period 1974 to 1979. The mill was expanded to 150 tons capacity in 1980. Operations were terminated and the plant closed in 1981.

In 1983, The Quinto Mining Corporation purchased the Chaput (Lumby) property and increased the property size. Geochemical and geophysical surveys outlined coincident geochemical and V.L.F.-E.M. anomalies near the top of Saddle Mountain. A trenching program followed which exposed the Plateau Shear Zone. Sampling of the shear zone produced encouraging gold and silver results from brecciated quartz veins in a graphitic host. During 1985, an initial reverse circulation drill program of 10 holes was completed and followed by 1,396 metres of diamond drilling in 13 holes. In 1986, the Saddle Mountain portion of the property was geologically mapped, additional ground V.L.F and magnetometer surveys







conducted and 2,700 metres of NQ diamond drilling completed along the Plateau Shear Zone. During 1987, 32 reverse circulation and 7 diamond drill holes for a combined total of 3,030 metres was completed. Additional ground V.L.F.-E.M., magnetometer and geochemical surveys were conducted. An initial metallurgical test was completed by Lakefield Research. In early 1988, Kilborn Engineering constructed a computer generated model of the Plateau Shear Zone based on 21 vertical sections.

The Quinto Mining Corporation entered into a joint venture agreement with Golden Seville Resources Ltd. in the spring of 1988. Purpose of this agreement was to test a proprietary column leach process concept. During the period July to November, 1988, Sancold Resources Contractors Inc. completed 186 metres (610 feet) of 3.05 x 3.66 metre (10 x 12 feet) exploration drift in the hanging wall of the Plateau Shear Zone and two cross-cuts totalling 105 metres (344 feet) of 2.74 x 3.05 metres (9 x 10 feet). A preliminary feasibility study was completed by Bechtel Canada, Inc. in December At this point, Golden Seville Resources Ltd. could not 1988. fulfil its financial commitments and the column leach testing project was terminated. The property was inactive during 1989. During 1990, M. Bradley (1990) geologically mapped and sampled the Plateau Shear Zone workings. The Quinto Mining Corporation initiated this work having noted not only the variable gold content (nugget effect) but also the presence of abundant graphite. Bradley recommended bulk sampling to evaluate the gold and graphite content over the entire width of the Plateau Shear Zone. To date, the Bradley (1990) recommendations have not been implemented.

The property was visited by the writer during the period of May 11 to 12, 1992 and February 3-5, 1993.

### REGIONAL GEOLOGY

Regional geology of the Lumby area is taken from Okulitch,(1979) and is shown on Figure 4. Modifications have been made to the geology map base for purposes of clarity and to reduce the number of rock units, some of which have no bearing on the presence or absence of mineralization.

Earlier geologic mapping by Jones (1959) of the Vernon map area shows the Lumby property mainly underlain by undefined rocks of the Shuswap Terrane-Monashee Group (Archean or later) and bracketed by two parallel northwest trending major faults.

Subsequent work by Okulitch,(1979) totally redefines this earlier interpretation. Jones (1959) considered the Shuswap Metamorphic Complex to be either an extension of the Archean and Proterozoic Canadian Shield (Okulitch, 1979). Okulitch (1979) has found that along the western margins of the Shuswap Metamorphic



### LEGEND FOR FIGURE 4 (Modified from Okulitch, 1979)

### MESOZOIC

### <u>Cretaceous</u>

Kg Granite, Granodiorite, Lesser Quartz Monzonite and Quartz Diorite.

#### Jurassic

Jgn Massive and foliated, syntectonic Pegmatite, Aplite, Leucocratic Granite, and Quartz Monzonite bordering and within Shuswap Metamorphic Complex and Okanagan Plutonic and Metamorphic Complex; Silver Star Intrusions: (May include Orthogneiss of Paleozoic and Proterozoic ages).

#### INTRUSIVE CONTACT

<u>Triassic and Jurassic</u> Nicola Group (Possibly includes Slocan Group)

### TRJnv

Andesite and Basalt flow rocks, porphyritic augite Andesite, Breccia, Tuff, Agglomerate, Greenstone, chloritic Phyllite, minor Argillite, Limestone, Sericite Schist.

Slocan Group Sicamous Formation

### uTRsc

Sericitic, graphitic and argillaceous Limestone, calcareous Phyllite, Argillite.

#### uTRsp

Shale, Argillite, massive Siltstone, Phyllite, Tuff and calcareous Pelite, minor Conglomerate, Limestone.

### uTRscg

Conglomerate.

Proterozoic and Paleozoic (may include Archean)

Shuswap Metamorphic Complex

#### PIPns

Undivided granitoid Gneiss, Paragneiss, Schist, minor Quartzite, Marble, Amphibolite.

#### PIPsb

Quartz mica Schist, commonly garnet and sillimanite bearing.

Complex, i.e., the Lumby area, units of probable Mesozoic and Paleozoic age can be traced into the Complex. Where this appears to be the case the units have been extracted from the Complex and correlated with those of known stratigraphic affinity (Okulitch, 1979).

Okulitch (1979) has mapped the Lumby area as being mainly underlain by shale, argillite, massive siltstone, conglomerate, phyllite, tuff, minor andalusite, staurolite and kyanite schist, limestone, greenstone and chloritic phyllite of the Slocan Assemblage. Okulitch (1979) could not locate the faulted boundary between the Slocan and Sicamous Formations and therefore has included everything in the Slocan Formation. The faulted block of Jones (1959) is not present. Okulitch (1979) shows the area covered by the Lumby property as being centred on a parallel westerly trending anticline-syncline structure that is parallel to the well defined Silver Star Anticline whose axis is just north of the property (See Figure 4).

#### PROPERTY GEOLOGY

Property geology for this report will be restricted to the Saddle Mountain portion of the property. Kuran (1985) mapped a hornblende diorite at the southern end of Saddle Mountain to the north of which occurs variably metamorphosed sedimentary rock and/or fine grained volcanic rocks. A sequence of argillaceous rocks lies interbedded with fine grained micaceous schists, the later being metamorphosed sedimentary rocks in the opinion of the writer rather than metamorphosed acid volcanic tuffs as suggested by Kuran (1985).

Of particular importance is the presence of the micaceous graphite-gold bearing Plateau Shear Zone. This shear zone strikes east-west with a dip of 45 degrees to the south toward the hornblende diorite intrusive and is the site of contorted quartzcarbonate(siderite)-pyrite veins bounded by massive foliated micaceous graphitic (mylonitic ?) schist containing pods of massive pyrite and scattered disseminated and sheared pyrite. The shear limits are bounded by harder, coarsely jointed, fine grained micaceous schist. The entire sequence appears to have been of a sedimentary nature originally. Through the mechanism of regional metamorphism and tilting during intrusion of the diorite, the less competent argillaceous beds became a plane of weakness along which movement occurred. This movement allowed the introduction of not only mineralizing solutions but also heat the result of which produced the quartz-pyrite-gold veins, the pyritic gold-bearing pods and the formation of graphite. A continuum of structural readjustments along this plane of weakness produced the contorted and foliated Plateau Shear Zone that is observed today. Kuran (1985) geology map is reproduced herein as Figure 5.

### GOLD MINERALIZATION - PLATEAU ZONE

Gold in the east-west striking, south dipping Plateau Shear to be associated with the abundant pyrite appears Zone The zone has been surface trenched, partially mineralization drilled along strike and down dip, intersected in underground working, metallurgically assayed and tested. Sulphide mineralization consists of abundant pyrite with minor amounts of galena, sphalerite and chalcopyrite. Marcasite has been mentioned Native gold occurs within and along fractures as a constituent. in pyrite and in quartz material. Microscopic examination of polished sections has indicated the presence of the mineral petzite - a gold, silver telluride. Non sulphide minerals include guartz, carbonate (ranging from calcite through ferroan dolomite to siderite) sericite or muscovite, clay altered feldspar and ultrafine graphite (See graphite investigation). Recent petrographic work suggests that the bulk of the so called "graphitic material" is in fact very fine grained muscovite/sericite interleaved with ultra fine grained graphite (0.1-0.3 by 3-5 microns) thus giving the rock its black appearance.

Prior analytical work {Kuran (1986 and 1987), Lebel (1987)} has not shown good reproducibility in gold content within the Plateau Shear Zone.

Kuran (1987) completed preliminary gold grade calculation on a longitudinal down dip section (reproduced herein as Figure 6). This preliminary work suggested a geological resource figure in the order of (1) 255,893 tons at an average grade of 0.115 opt Au; or (2) 334,685 tons at an average grade of 0.102 opt Au; or (3) 518,266 tons at an average grade of 0.075 opt Au.

The main point of the above was to indicate that gold is noted throughout the Plateau Shear Zone where utilized intercepts ranged from 0.006 opt Au to 0.226 opt Au.

The above calculations were based in part on geochemical analytical methods which are subject to error due to the small sample size used in this method. Samples from percussion holes RC-85 - 1 through 11 and diamond drill holes 85 - 2 through 9, 86 -1 through 18, 86 ~ 18, 86 - 22, 86-33, and 87 - 1 were analyzed by rock geochemical methods (i.e. on a 10 gram sample). The rest of the 87 reverse circulation drill program used a 500 gram split at minus 10 mesh which was pulverized to minus 150 mesh before a 30 gram sample (one assay ton) was taken.

In the recent investigation, 22 chip/channel samples (minimum 5 kilograms per sample; Sample width mostly 2 metres) were



# LEGEND FOR KURAN (1987) GEOLOGY MAP - FIGURE 5

#### TABLE 2

#### Table of Formations

10 Malic Dykes

9

5

3

1

Hornblende Gabbro; medium to coarse crystalline, granular, dark green. 10. 10b Hornblendite; very line grain, dark green to black.

-- Intrusive Contact --

Biotite Granite Very line to line crystalline, 5-13% line pyrite, light grey.

-- Intrusive Contact --

Hornblende Feldspar Porphyry Dyke 1 Light grey, fine crystalline, hornblende lathes.

- Intrusive Contact --

7 Polyphase Hornblende Diorite Undifferentiated very line chilled to coarse crystalline, hornblende or feldspar porphyritic, weakly foliated, variably pyritic.

-- Intrusive Contact --

4 Argillite

- Well bedded to laminated, black, interbeds of fine grey felsic ash, variably calcareous.
- Blotite Siltstone
  - Sa Fine grained, moderately to poorly sorted, well bedded, minor argillite, beds of argillite rip-up clasts.
  - Turbeditic cycles, fine siltstone to coarse pebbles, graded 36 into unit 4b.
- Intermediate Volcanic Tulls
  - Ash tull; line to medium grain, drab green, massive. 42
  - 4Ь Lithic tuff; poorly sorted, turbeditic cyclic members,
  - polymictic matrix supported.
  - Argillaeous ash tull; fine grained, finely bedded. 4c
- Acid Volcanic Tulls
  - Ash Tuft; fine grain, rare lapilli, ligh grey, locally sericitic rich and crenulated, minor disseminated pyrite. 3.
  - Lapilli-ash tuffs; ash to medium lapilli size, dacite to andi-dacite 36 composition, variably calcareous.
  - Crystal tull; line sericitic ash matrix, 15% 0.3-2 mm subrounded Jc feldspar or quartz crystals.
  - Spotted-banded tull; calcareous, greenish grey, knots of brown 3d biotite and pyrrhotite.
- 2 Argillite / Shear Zone
  - 22 Argillite; moderately to intensely sheared, graphitic gouge, pyritized, minor quartz velning.
  - Quartz veins; sugary to bullish, variably pyritic, up to 60% pyrite, 2ь minor pyrrhotite, trace galena and sphalerite. Argillite; poorly sheared, 2-3% 1 cm buckshot pyrite cubes.
  - 2c
    - Acid Volcanic Tulls
      - Crystal tulf; dark grey line felsic matrix, 10-20% 0.5-1 cm euhedral feldspar crystals, minor subrounded quartz eyes.

collected by the writer from the underground intersection of the Plateau Shear Zone. All samples were carefully collected to reduce any sampling error that may have been inadvertently introduced in previous sampling. All assay results (Appendix A) were reported using a one assay ton analysis. Figure 7 (in pocket) shows the location and values of the recent sampling. The 190 X-cut based on 12 samples over a 20 metre width averaged 0.044 O.P.T. gold with a range of 0.001 to 0.313 O.P.T. and the 140 X-cut based on 11 samples over a 14 metre width averaged 0.060 O.P.T. gold with a range of 0.005 to 0.220 O.P.T. gold. The overall average of both cross cuts is 0.052 O.P.T. gold. A comparison was made with previous sampling (Kuran, 1986, 1987) and it was obvious that the most recent sampling and/or analysis still has not addressed the analytical repeatability problem, particulary in light of past and present metallurgical testing.

In comparing the smaller X-cut samples with the metallurgical results (See following discussion on metallurgy) which were all based on much larger bulk samples, the gold distribution appears to have good distribution and be of higher grade. The smaller samples from drill intersections and wall sampling have very erratic distributions with lower grade than the larger samples used in the metallurgical tests.

### **GRAPHITE INVESTIGATION**

Previous investigations by the writer and A.D. Drummond, Ph.D., P.Eng. lead to the conclusion that the Plateau Shear Zone contained a very high percentage of graphite due to the dense black sheen exhibited by the material in the zone. The black material behaved like graphite including its ability to mark paper. Subsequent assaying by the classical "Leco" method indicated that the rock only contained a trace of graphite. This inconsistency was finally addressed by changing the assay procedure. The "classical leco method" involves pre-heating the sample to 600 degrees centigrade to remove any organic carbon and/or carbonate carbon prior to placing it in the Leco Analyzer. The new method involved using a nitric acid wash to remove any organics, sulphur, calcium carbonate and soluble iron followed by a hydrofluoric acid wash to remove the silica. The sample was then analyzed by the "leco" method without the pre-heat step to determine carbon A comparison of the two methods is shown in Table 1 content. below: Assay certificates are in Appendix A.



### FIGURE 6

LONGITUDINAL DOWN DIP SECTION

PLATEAU SHEAR ZONE

THE QUINTO MINING CORPORATION

(After Kuran (1987), Figure 34)

GOLD VALUES ARE IN OUNCES PER SHORT TON

(17)

### TABLE 1

### A comparison of carbon assay methods

Sample No. 5015B

Method 1	Method 2
Preheat oxidation/leco	Wet chemical oxidation/leco
Graphite carbon percent	Graphite carbon percent
0.23	5.35

Sample 5015B is a flotation product produced by Westcoast Minerals Testings Inc. (See Appendix A and section on metallurgical testing) from a composite sample of fresh graphitic material from the underground workings.

The above discrepancy in carbon content between the two assay methods is explained by the fact that the oxidation rate (i.e. burning) is related to surface area of the mineral/element particle. In the present case it is now known that the graphite in the Plateau Zone samples is extremely fine grained. Under 1000 magnification it was determined that most of the graphite grains were in the 3-5 by 0.1-0.3 micron range which translates into a maximum surface area for each grain. Therefore in the thermal oxidation/leco method, the pre-heating to 600 degrees centigrade of the sample not only drove off the organic and carbonate carbon but also most of the fine grained graphite carbon prior to putting it in the Leco Analyzer. The thermal oxidation of very fined grained graphite was confirmed by submitting two chemically pure graphite samples for analysis using the "Classical Leco" method. The two samples (KS-75 and No. 763) were very fine grained (approx. 325 mesh) and were guaranteed to contain 99.9 and 96 percent carbon respectively. As can be seen in the following Table 2 the assay results using the "Classical Leco" method does not even approximate the correct value ( See Appendix A).

### TABLE 2

### Classical Leco assays of graphite standards

Sa (S	ample KS-7 99.9% grap	5 hite)					Sample (96%	No. 763 carbon)
L.O.I.	TOT/C	GRA/C				L.O.I	. <b>T</b> OT/C	GRA/C
91.5%	100.49%	38.16%				91.6%	89.39	8 7.37%
L.O.I.	= Loss on	ignition,	TOT/C	=	Total	carbon,	GRA/C =	Graphite

### X RAY DIFFRACTION INVESTIGATION

Flotation product F-1 (Sample 5015B) was analyzed by A.D. Drummond on a Phillips X-Ray diffractometer at the University of British Columbia, Department of Metals and Material Engineering (See Appendix B for X-ray patterns). The purpose of this investigation was to determine the mineralogical content of the flotation product that looked like graphite but did not analyze as graphite using the LECO technique. On November 18, 1992 and on January 13, 1993, XRD patterns were obtained using copper radiation with a nickel filter, 36 Kv and 20 ma. "Graphitic material" from the light fraction of a Falcon concentrator and "graphitic material" obtained from the slip surface of graphitic schist gave similar patterns with d-spacings that indicated the presence of quartz (ASTM 33-1161), muscovite (ASTM 6-263) and graphite (ASTM 25-284) all of which have their major peak at 3.330 to 3.324, minor constituents were a chlorite phase at 14.12 Angstroms (ASTM 39-381), calcite (ASTM 5-586), possible siderite (ASTM 29-696) and pyrite which was megascopically identified but was not present in sufficient quantity to be detected on the diffractometer pattern.

The next part of the X-ray diffraction investigation used two graphite standards for comparison to a flotation concentrate (F-1) in which the flotation product was obtained using only Dow Froth produced 250. Graphite floats easily and the product megascopically appeared to be monomineralic with the exception of The standards referred to above were (1) a minor pyrite. commercial graphite known as Dixon KS-75 (99.9% synthetic carbon graphite) and (2) another commercial carbon product known as Asbury 763 (96% carbon composed of about 50% graphite and 50% lamp black). The KS-75 sample produced an excellent diffractometer pattern with the major d-spacing at 3.41 Angstroms and a good correlation to ASTM 41-1487. The Asbury 763 pattern was very different in that

(18)

the main peak was more diffuse between 3.411 and 3.437 Angstroms with almost an amorphous build up pattern between 23.0 and 26.3 degrees Two-Theta. The Asbury 763 pattern also indicated the very minor presence of muscovite (9.93 Angstroms), talc(?) (9.3 Angstroms) and kaolinite(?) (7.07 Angstroms). The flotation product F-1 indicated peaks for the presence of muscovite and graphite with minor pyrite and quartz and a possible trace of calcite (very low peak at 3.025 Angstroms which is the major peak for calcite). See Appendix C for X-ray patterns.

From the above, it is apparent that the megascopically uniform black flotation product F-1 contains mostly muscovite with a major constituent of graphite and minor to trace amount of pyrite, quartz and probably calcite.

### METALLURGY

Prior to the present re-sampling and metallurgical testing a comparison with previous metallurgical work by Lakefield Research (Salter, Sarbutt and Rollwagen, 1987) was made in an attempt to isolate the repeatability problem. The results of these tests are summarized as follows:

Two bulk samples were taken from the Plateau Shear Zone and investigated for gold recovery by Lakefield Research (Salter, Sarbutt and Rollwagen (1987) and summarized by Kuran (1987) and by Richards (1988).

#### A. SURFACE VERSUS BULK GOLD GRADE

One sample was composed of reject from RC holes 85-1 to 10 and referred to as "core sample" in the reports. It is important to note the following:

(1) weighted grade from drill assays = 0.092 opt gold and (2) calculated head assay = 0.152 opt gold.

This is a significant increase in the assigned gold value to the "core sample".

The second bulk sample was taken from the West Pit area and is referred to as "West Pit" or "Trench sample" (oxidized material). Again it is important to note the following:

- (1) weighted average grade
  - of chips from same area = 0.109 opt gold, and as "Trench sample"
- (2) calculated head assay = 0.27 opt gold.

As in the above, this is a significant increase in the assigned gold value to the "Trench sample".

### B. GRAVITY AND GRAVITY-FLOTATION TEST

Gravity - Flotation testwork suggested that gravity separation alone gave gold recoveries of 18.3% for the "Trench sample" and 21.9% for the "core sample". Gravity combined with rougher flotation gave gold recoveries of 80.6% for the "Trench sample" and 98.3% for the "core sample".

### C. CYANIDE LEACH TESTS

Cyanide agitation leach tests on finely ground material (+95% passing 200 mesh) suggested 85.9% gold recovery in 48 hours on the "core sample" and 88.7% gold recovery on the "Trench sample". When these tests were done with carbon-in-leach, the results in 48 hours were 95.4% gold recovery on the "core sample" and 97.4% gold recovery on the "Trench sample".

The gold recovery with time is shown below for the tests done with carbon-in-leach: Recovery to carbon Residue Head Grade Sample g/t opt 24 hrs 6 hrs 48 hrs g/t opt ጽ 8 Au 8 Au Au Au "core" 85.9 3.40 0.099 0.48 0.014 "core" with 85.9 94.3 95.4 3.64 0.105 0.16 0.005 carbon "Trench" 1.10 0.032 88.7 9.76 0.283 "Trench" 96.9 97.4 with 0.27 0.008 89.3 10.70 0.310 carbon

The above indicates that gold is readily solubilized with cyanide within 24 hours (NOTE: recovery in 6 hours).

The presence of a telluride in the testwork is not mentioned.

The present on going metallurgical testing by Westcoast Mineral Testing Inc. was originally initiated to define the possibility of recovering graphite in economic quantities as a by product of the gold mineralization (Hawthorn, 12/23/92; in Appendix C). The results of the initial investigation showed that graphite was present but in a form that would probably not be recoverable as a separate product (See "GRAPHITE INVESTIGATION" above).

In the present investigation approximately 10 kilograms of graphitic underground material was dry crushed to 90 percent passing 20 mesh by the Placer Research Centre. Westcoast Mineral Testing Inc. combined this material with higher percentage pyritic material to approximate "mine run" rock. A summary of the Westcoast Mineral Testing Inc. test is as follows (From Hawthorn, 1/3/93):

### SAMPLE COMPOSITION

The test sample consisted of 73 percent graphitic material assaying 0.024 O.P.T. gold and 27 percent pyritic material assaying 0.60 O.P.T. gold for a combined head assay of 0.180 O.P.T. gold.

#### PROCEDURE

Grind 1000 gm / 5 min. / 67% solids Pan: K = +200:1 Float: Staged rougher to completion Wet/dry screen rougher tailings to 325 mesh

### METALLURGICAL CALCULATIONS

			As	зау	Di	stribution	b.
Product	Wt%	Au oz./t	LOI%	Fe%	5%	Au	
Pan conc.	1.1	2.28				9.1	
F-1 RC F-2 RC	<b>15.2</b> 20.9	0.136 0.980	11.3 25.1	5.5 34.4	2.3 35.6	7.6 74.6	
F	36.1					82.2	-
Tails	62.8	0.038				8.7	
Feed(calc)	100.0 Assav	0.275				100.0	•

### RC = Rougher Concentrate

Note when comparing the results of the latest test above it again shows a very significant increase in gold content in comparison to the head assay.

The F-1 product although looking like pure graphite only contained 5.35 percent graphite carbon. Visually under high magnification (60x) fine grained pyrite (4-5 percent) was visible which explains the above iron and sulphur assays. A whole rock analysis of the F-1 product (Sample 5015B, Appendix A) confirmed that one or more silicate minerals constituted approximately 90 percent of the product. The percentages of silica, aluminum and potassium were consistent with a muscovite/sericite composition as indicated by the X.R.D work (See X-ray Diffraction Investigation).

Preliminary petrographic analysis suggests that the graphitic portion of the host rock is mainly composed of very fine grained muscovite/sericite. The reason that it appears so readily floatable is the ultra fine grained nature of the interleaved graphite which makes the muscovite grains behave like graphite. Muscovite is not normally floatable with low concentration of DF 250 (Dow Froth 250). It should be noted that in excess of 15 percent by weight of the original sample was a flotation product. If a pure enough muscovite product can be produced using this natural floatable characteristic it is very likely that a "By Product" market can be found since fine grained muscovite has numerous industrial applications. Future metallurgical investigations will be directed toward producing a saleable product of this nature.

### CONCLUSIONS AND RECOMMENDATIONS

The 1992-1993 exploration program on The Quinto Mining Corporation Lumby property has concentrated on a better definition of the gold grades of the previously defined Plateau Shear Zone and to investigate the possibility of additional "By Product" values, i.e., Industrial Minerals associated with the shear zone. To this end the present program consisted of re-sampling a portion of the underground working, assaying, conducting mineralogical and metallurgical testing.

Early indications suggested that a valuable graphite resource existed in the Plateau Shear Zone, but subsequent tests proved that graphite although present, is not in a readily extractable form, i.e., too fine grained and too tightly bound to muscovite. The graphite occurs as ultra fine grained grains interleaved in very fine grained muscovite/sericite grains. The graphite causes the muscovite/sericite to be readily floatable which in turn may have a considerable "By Product" value.

Sampling of the walls of the cross cuts has produced variable gold values and have indicated that the gold is not necessarily associated with the quartz as was originally thought. Sampling of the underground workings has suggested that there is free gold as well as gold with pyrite which can occur within the quartz vein material, in graphitic material adjacent to the quartz vein within the pyrite boudins and disseminated pyrite material, throughout the shear zone. Recent sampling and assaying has yet to address the previous repeatability and actual grade problem. It still can be shown that more gold is recovered using large samples and metallurgical methods (flotation or barrel leach using cyanide) rather than the small samples used for standard fire assay (small chip/channel samples). It is probable that the actual gold content of the Plateau Shear Zone can only be defined by using bulk sampling techniques.

Based on the success to date in defining better gold grades when large samples are analyzed and the presence of what appears as a large quantity of "By Product" muscovite and/or a graphite/muscovite product it is recommended the following investigations be instituted:

- (1) Conduct further mineralogical studies to better define the character and composition of the fine grained muscovite in the zone.
- (2) Continue with the metallurgical testing to optimise the recovery of the muscovite - muscovite/graphite product with special attention given to producing a product with ASTM specifications.
- (3) Conduct metallurgical tests on the leachability of the graphitic material with an additional emphasis on the depression of pyrite with cyanide. These tests will hopefully define a higher gold grade for the deposit than what conventional assaying has indicated.
- (4) Conduct metallurgical tests on the low temperature removal of graphite from the graphite/muscovite. In other words produce a saleable white muscovite product.
- (5) If all or most of the above tests are successful then a program of bulk sampling should follow to develop a flow sheet for the deposit.

### PERSONNEL TIME DISTRIBUTION

(B.S 1-5, P.S., P.S. 2-4, 7, Quin, M.M. 1-5, Lum 1-4)

- A.D. Drummond, Ph.D., P.Eng. (Geological Engineer)
  - Field May 11-13, 1992 inclusive 3 days
  - Office Research and report preparation 120 hours Period January 5 - February 28,1993
- D.A. Howard, M.Sc., P.Eng. (Geological Engineer)
  - Field May 11-13, 1992 inclusive February 3-5, 1993 inclusive 6 days
  - Office Research and report preparation 100 hours Period January 5 - February 28, 1993
- Gary Hawthorn, P.Eng. (Metallurgist) Time included in Westcaost Mineral Testing Inc. invoice
- Steve McAlister, P.Eng. (Metallurgist) Time included in Falcon Concentrators Inc. invoice

(24)

Jerry White (Sampler/Miner)

6 days

Bernhard Klein, Ph.D. (Senior Process Metallurgist) Time included in Process Research Associates Ltd. invoice

Cat Operator Time included in Chaput Logging invoice.

### COST STATEMENT

### PERSONNEL

A.D. Drummond, Ph.D., P.Eng. Field 3 days @ \$400/day Office 120 hours @ \$50/h	r. (Research)	\$1200.00 6000.00
D.A. Howard, M.Sc., P.Eng. Field 6 days @ \$400/day Office 100 hours @ \$50/h	r. (Research & Report)	2400.00 5000.00
Jerry White, Sampler/Miner Field 6 days @ \$250/day		1500.00
	Sub-total GST @ 7%	16,100.00 1127.00
	Sub-total	\$17,227.00
EXPENSES AND DISBURSEMENTS		
Westcoast Mineral Testing Inc. Metallurgical Testing		\$ 584.00
Falcon Concentrators Inc. Metallurgical Testing		500.00
Process Research Associates Metallurgical Testing		375.00
Acme Analytical Laboratories Ltd Assaying	l.	93.00
Eco-Tech Laboratories Ltd. Assaying		370.00

The University of British Columbia Metals and Materials Engineering Depart X.R.D. analysis	ment	74.00
The University of British Columbia Department of Geological Sciences Optical and microprobe analysis		800.00
Cominco Research Laboratory Petrography		134.00
Chaput Logging Cat work opening road (Snow removal)		1338.00
Transportation 4x4 truck - 6 days @ \$100/day		600.00
Accommodation and food 2 rms., 4 nights in Vernon		820.00
Report preparation (Typing and copying)		100.00
	Sub-total	5788.00
	TOTAL	\$23,015.00

Respectfully submitted, Mar 9/93 A. D. DRUMMOND Munn A.D. Drummond, Ph. Bress Eng e a a a ROVI DAVID A. HOWARD D,A. Howard, M.Scorphin Eng.

### CERTIFICATION

I, Arthur Darryl Drummond of the City of Vancouver, Province of British Columbia, hereby certify as follows:

- 1. I am a geological engineer residing at 3249 West 35th Ave., Vancouver, B.C., V6N 2M9 and employed by D.D.H. Geomanagement Ltd., with an office at 422 - 470 Granville St., Vancouver, B.C., V6C 1V5.
- 2. I am a registered Professional Engineer of the Province of British Columbia, certificate number 5778. I graduated from the University of British Columbia in 1959 with a B.A.Sc. in geological engineering, and in 1961 with a M.A.Sc. in geological engineering. I graduated from the University of California at Berkeley in 1966 with a Ph.D. in geology.
- 3. I have practised my profession continuously for 30 years primarily with the Placer Development Group of Companies at Craigmont, Endako and Gibraltar mines, and in mineral exploration in Canada, United States of America, Chile, Argentina, Mexico and the Philippines.
- 4. I am a co-author of this report which is based on personal supervision of the described work program and from data contained in the files of D.D.H. Geomanagement Ltd., private reports and government publications.
- 5. I hold an employee stock option on shares of The Quinto Mining Corporation.
- 6. This report may be utilized for development of the property provided that no portion may be used out of context in such a manner as to convey a meaning which differs from that set out in the whole.
- 7. Consent is hereby given to The Quinto Mining Corporation to use or reproduce this report or any part of it for the purposes of development of the property, or related to the raising of funds.

Dated at Vancouver, B.C. this 9th day of March, 1993.

n.

A. D. DRUMMOND

A.D. Drummond, Ph.D., P.Eng. D.D.H. GEOMANAGEMENT LTD.

### (26)

# (27)

# CERTIFICATION

I, David A. Howard, of the City of Vancouver, Province of British Columbia, hereby certify as follows:

- I am a geologist residing at 9040 Glenallan Gate, Richmond, B.C., with an office at 422-470 Granville Street, Vancouver, B.C.
- 2. I am a registered Professional Engineer of the Province of British Columbia. I graduated from Montana State University in 1964 and from the University of Washington in 1967.
- 3. I have practised my profession continuously since June, 1966.
- 4. I am a co-author of this report which is based on personal supervision of the described work program and from data contained in the files of D.D.H. Geomanagement Ltd., private reports and government publications.
- 5. I hold an employee stock option on shares of The Quinto Mining Corporation.
- 6. This report may be utilized for development of the property provided that no portion may be used out of context in such a manner as to convey a meaning which differs from that set out in the whole.
- 7. Consent is hereby given to The Quinto Mining Corporation to use or reproduce this report or any part of it for the purposes of development of the property, or related to the raising of funds.

Dated at Vancouver, B.C. this 9th day of March, 1993.

DAVID A. HOWAR

David A. Howard, M.Sc., P.Eng.

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- Bradley, M. (1990) Report On Chip/Channel Sampling And Geological Mapping Of The 140East And 190East Crosscuts, 808 m Level Underground, Chaput 5 Claim: a private report for The Quinto Mining Corporation by Mike Bradley and Associates, dated October 1, 1990.
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# APPENDIX A

# ASSAY CERTIFICATES

PHONE (604) 253-3158 FAX (604) 253-1716 ACME ANALYTICAL LABORATORIES LTD. 852 E. HASTINGS ST. VANCOUVER B.C. V6A 1R6 ASSAY CE FICATE D.D.H. Geomanagement Ltd. File # 93-0027R GRA/C SAMPLE# 5015B 5.35 

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14	- 27128	7.56	.220	6.0	.18	
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	250g TAILS	19	49	36	269	9.0	57	8	660	3,16	27	5	ND	3	261	4.7	3	2	46	3.13	.051	8	62	1.61	165	.01	3	2.10	,02	.32	
	750g TAILS	16	42	26	234	1.4	51	6	618	3.05	-23	5	ND	2	249	3.9	2	2	- 43	2.95	.037	<u>6</u>	55	1,68	80	01	2	2.14	.01	.29	
	67003	12	93	35	645	1.9	118	30	813	6.19	143	5	ND	1	306	11:4	2	2	24	4,11	.091	2	- 55	1.32	23	.01	, 4	1.05	.01	.14	
1.4	69074	12	91	35	631	6.7	195	29	797	6.17	139	5	ND	1	297	11.2	2	2	25	4.05	.090	§ 3	- 54	1.30	51	.01	- 4	1.08	_01	- 16	220
	RE 67003	11	87	28	640	1.8	- 99	- 28	792	6.06	138	5	NÐ	1	295	11.0	2	2	23	4.07	.090	3 2	53	1.28	18	.01	5	.96	-01	-13	127
																						<u>}</u>				<u> (2005)</u>	· _,				1021
	STANDARD C	18	60	39	129	7.2	67	31	1043	3.96	- 42	18	7	- 37	- 53	18.0	14	- 21	56	.50	.087	8 <b>39</b>	60	94	188	S 97	: 34	1.85	.00	- 14	(1, 1)

ICP - .500 GRAM SAMPLE IS DIGESTED WITH 3ML 3-1-2 HCL-HN03-H20 AT 95 DEG. C FOR ONE HOUR AND IS DILUTED TO 10 NL WITH WATER. THIS LEACH IS PARTIAL FOR WW FE SR CA P LA CR MG BA TI B W AND LIMITED FOR WA K AND AL. AU DETECTION LIMIT BY ICP IS 3 PPM. DATE RECEIVED: DEC 15 1992 DATE REPORT MAILED: Dec 12/92 SIGNED BY....D.TOYE, C.LEONG, J.WANG; CERTIFIED B.C. ASSAYERS

TOTAL P.002

# APPENDIX B

# X-RAY DIFFRACTION PATTERNS











FOR QUINTE MINING CORP.



5- 586	JCP	DS-ICDD Copyright	(c) 1991		Qualit	y: i					Ca	late.
CaCO 3							2 <i>€</i>	29.	4	d A 3.86 3.035	1 Int.	h k 1 0   2 1 0 4
Calcite, sy	rbonate /A		Filter			 d-		- ,,		2.845 2.495 2.285	3 14 18-	0 0 6 1 1 0 1 1 3
Radi Culai Cutoff: Ref: Svanso	Int Int Int En, Fuyat	: Diffractometer , Natl. Bur. Stan	1/1cor d. (U.S.), (	2.00 Circ. S	i39, II	51 ( 	(1953)			2.095 1.927 1.913 -1.875	18 5 17 17	2 0 2 0 2 4 0 1 8 1 1 6
Sys: Rhombo a: 4.989 A: Ref: Ibid. Dy: 2.71	ohedral ( b Ba:	Hex) S.G.: R- : : 2.71 SS/FOM:	3c (167) c: 17.062 C: F30≠50(.016	.37)	A: Z: 6		C: mp:	3.4199		1.626 1.604 1.587 1.525 1.518	4 8 2 5 4	2 1 1 1 2 2 1 0 10 2 1 4 2 0 0
ea: 1.487 Ref: Dana	nv8 s System	1: 1.659, ey: of Mineralogy, 7t	h Ed., 2 14	Sign: 2	- 2	۷:	•••••			1.510 1.473 1.440 1.422	3 2 5 3	1 1 9 1 2 5 3 0 0 0 12 2 1 7
Color: Cole X-ray patte 13397-26-7. Cu, X, Mg, 190. Other Not permit PSC: hR10.	orless ern at 26 . Spectr Ma, Si, r form: a tted by s Mwt: 10	C. Sample from oscopic analysis: Sn; (0.00011 Ag, iragonite. Calcit pace group. 0.09. Volume(CD)	Mallinckrod 〈O.II Sr; Cr, Fe, Li, e group, ca ; 367.78.	t Chemi (0.017 Mn. 1 lcite s	ical Wor Ba; <0. Ierck In Subgroup	ks. 0011 dex,	CAS ( Al, , 8th	no.: B, Cs, Ed., p	•	1.339 1.297 1.284 1.247 1.235 1.1795	2 2 1 2 3	0 2 10 1 2 8 3 0 6 2 2 0 1 1 12 2 1 10
d A L	 	• k l	   d A	[nt.		 h	k 1	•••••	: d A	[nt.		
1.1538 1.1425 1.1244 1.0613 1.0473	3 1 (1 1 3	1 3 4 2 2 6 1 2 11 2 0 14 4 0 4	0.9846 0.9782 0.9767 0.9655 0.9636	l 1 3 2 4	ſ	2 1 3 4	3 2 3 10 2 14 2 4 0 8	   	+			
1.0447 1.0352 1.0234 1.0118 0.9895	4 2 (1 2 (1	3 1 8 1 0 16 2 1 13 3 0 12 3 2 1	0.9562 0.9429 0.9376	<1 2 2		2 4 2	0 16 1 0 2 12	•				

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Strong lines: 3.04/X 2.29/2 2.10/2 1.91/2 1.88/2 2.50/1 3.86/1 1.60/1

39-381 JCPDS-ICDD Copyright (c) 1991 Quality:	Chlorite Group.
Na Al (Si,Al) O (OH) !H O O.5 6 B 20 10 2 Sodium Aluminum Silicate Hydroxide Hydrate	d A Int. h k 1 14.2 10 0 0 1 7.7 50 0 0 2 4.78 10 0 0 3 4.44 50 0 2 1 1
Chlorite-vermiculite-montmorillonite Rad: CuKa Lambda: 1.542 Filter: Mono. d-sp: Diff. Cutoff: 40.0 Int: Diffractometer I/Icor: Ref: Bayliss, P., James, D., Clay Miner., 16 213 (1981)	3.50     100     0     0     4       2.88     6b     0     5       2.56     15     2     0       2.50     15     1     3        2.34     20b     0     6
Sys:         S.G.:           a: 5.1         b: 8.9         c: 14.4         A:         C:           A:         B:         C:         I:         mp:           Ref:	1.98 / 6b 0 0 7 1.82 / 4b 0 0 8 1.66 / 6b 0 9
Dx: Dm: SS/FOM: ea: nvB: ey: Sign: 2V: Ref:	
Specimen from oil sand deposits, Alberta, Canada. A di/dioctahedral chlorite-vermiculite-montmorillonite irregular mixed-layer. 001 is 15.0 A glycollated, 12.4 A heated 400 C, 11.8 A heated 600 C. Pseudohexagoanl. Mixed-layer group, random subgroup. #Asymmetrical. See original PDF Card for Graphical diffractometer trace.	missing move
	j devet V

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Strong lines: 3.50/X 7.70/5 4.44/5 2.34/2 2.56/2 2.50/2 14.2/1 4.78/1

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25-284	JCPDS-1CDC	) Copyright (c) 1991	Qualit	y: C	Ana	phi	te -
C Carbon Graphite, syn				2t 26.6	d A 3.348 2.127 2.027 1.795	lat. 100 - 3 15 3	$ \begin{array}{c} h k \\ 2 \\ 24.6 \\ 41.6 \\ 1 \\ 0 \\ 47.7 \\ 1 \\ 1 \\ 0 \\ 4 \\ 1 \\ 0 \\ 4 \\ 1 \\ 0 \\ 4 \\ 1 \\ 0 \\ 4 \\ 1 \\ 0 \\ 4 \\ 1 \\ 0 \\ 4 \\ 1 \\ 0 \\ 4 \\ 1 \\ 0 \\ 4 \\ 1 \\ 0 \\ 4 \\ 1 \\ 0 \\ 4 \\ 1 \\ 0 \\ 4 \\ 1 \\ 0 \\ 1 \\ 0 \\ 4 \\ 1 \\ 0 \\ 1 \\ 0 \\ 1 \\ 0 \\ 1 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0$
Rad: CuKa Cutoff: Ref: Holcombe, Communication,	Lambda: 1. Int: Calcu USAEC Oak f (1974)	54178 Filter: Dated I/lcor: Ridge Y-12 Plant, Report	t Y1887 (1973	d-sp: Calculated 13, Private	1.5398 1.3154 1.2280 1.1529	4 1 4 6	60.0° 1 0 3 1 0 4 77.7° 1 1 0 83.8° 1 1 2
Sys: Hexagonal a: 2,456 A: Ref: 1bid. Dx: 2,28	b: B; D <b>a</b> :	S.G.: P63/mmt (194) c: 6.696 C: SS/FOM: F19=280(.003	A: Z: 4	C: 2.7264 mp;	1.1333 1.1160 1.0503 0.9902 0.9601		1 0 5 0 0 6 2 0 1 1 1 4 2 0 3 1 0 7
ea: Ref:	nvB:	ey:	Sign: 2	2V:	0.8370		008
Peak height in deleted by 2-5 Volume[CD]: 34	itensities. 25; good ex 1.98.	CAS no.: 7440-44-0. C periment pattern; Bayli	type. PSC: ss 6/90. Mut	hP4. To be 1: 12.01.	ŏ.7982	2	211
••••••••••••••••••••••••••••••••••••••		*******					

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Strong lines: 3.35/X 2.03/2 1.67/1 1.15/1 1.54/1 1.23/1 0.99/1 0.83/1

C Carbon	d A 3.38 2.139 2.039	1 Int.	20 <sup>°</sup> h k l
	1.807	6	42.2° 1 0 0 44.4 1 0 1 0 2
Graphite-2H Rad: CuKal Lambda: 1.54051 Filter: Ni d-sp: Diff. Cutoff: 22.1 Int: Diffractometer I/Icor: 7.78 Ref: Sanc, I., Polytechna, Foreign Trade Corporation, Panska, Czechoslovakia, JCPDS Grant-in-Aid Report, (1990)	1.681 1.548 1.234 1.1604 1.1208	1 3 3 (1	54.5 0 0 4 59.65 1 0 3 77.2 1 1 0 83.2 1 1 2 0 0 6 2 0 1
Sys: Hexagonal       S.G.: P63/mmc (194)         a: 2.4704(15)       b:       c: 6.7244(38)       A:       C: 2.7220         A:       B:       C:       I       4       mp:         Ref: Ibid.       Dr:       2.16       SS/FDM: F10=18(.042,13)       SS/FDM: F10=18(.042,13)	1.0367		
ea: nwB: ey: Sign: 2V: Ref:			
Color: Black Pattern taken at 25(1) C. Specimen from Netolice, Czechoslovakia. CAS no.: 7782-42-5. sigma(lobs)=+/-0.05. C type. Also called: cliftonite. Also called: plumbago. Silicon used as external standard. PSC: hP4. To replace 23-64 and 34-567, and validated by calculated pattern 25-284. Structure reference: Aust. J. Chem., 42 479 (1989). Mwt: 12.01. Volume(CD1: 35.54.			

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Strong lines: 3.38/X 2.04/1 1.68/1 1.23/1 1.16/1 2.14/1 1.81/1 1.55/1

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4-164		JCPDS-ICDD Copyright	; (c) 1991		Quality: i	ł	Kao	lià <sup>-</sup>	ite.	
Al_Si_O_	 (OH)						d A	Int.	h k 1	
Z Z Ə Aluminum Kaolinite	9 Silicat 2-1A	e Hydroxide					7.17 4.478 4.366 4.186	100 35 60 45	0 0 0 2 -1 1 -1 -1	1 0 0 1
Rad: CuKa Cutoff: Ref: Good	a dyear, D	Lambda: 1.5410 Int: Visual uffin, Mineral. Mag.	Filter 1/Icor , 32 902 (1	: Mono. 961)	d-sp:		4.139 3.847 3.745 3.579	40 40 25 80	0 -2 0 2 0 0	1 1 2
Sys: Tric a: 5.155	clinic	S.G.: P	·1 (2) c: 7.407		 А: С:		3,420 3,376	5 35	1 -1	1
A: 91.68 Ref: Ibid Dx: 2.60	s. D Da	B: 104.9 ; 2.64 SS/FOM:	C: 89.94 F30=19(.022	72)	l: 2 mp;		3.155 3.107 2.754 2.566	20 20 20 35 25	-1 -1 -1 -1 0 -2 -2 -2 -2 1 -3	2 2 1 0
ea: 1.559 Ref: Deer	9(6), r, Howie	nvB: 1.564(5), ey: , Zussman, Rock Fori	1.565(5), iing Mineral	Sign: s, 3 19	-, 2V: 24-50 deg. 4		2.535	35 10	-1 -3 1 -1	1
Color: Wb Specimen	hite som from Sc	etimes with reddish, alby, Yorkshire, En	brownish o land. Vali	r bluis dated t	h tints y calculated pattern B	org	2,495 2,385 2,347	45 25 40	2 0 0 0 -2 0	0 3 2
and Smith C.D. Cell a/b=0.826 Volume(CC	h, GSA M 1: a=7.4 58, c/b= D1: 330.	emoir, 122, Kaolin; 07, b=8.959, c=5.153 0.5754. PSC: aP34. 43.	i, alpha=90. To replace	ne grou 06, bet 12-447	p, dioctanedral subgru a=104.90, ga <b>nn</b> a=88.32, and 5-143. Mwt: 258.	16.	2.338 2.305 2.293 2.253 2.253	40 5 35 20	1 -3 -1 1 1 3 -1 -3 0 4	1 3 1 2 0
d A l	: Int.1	<u> </u>	i d A	Int.	h k l	l d A	¦ Int.¦		hki	
2.218 2.197 2.186 2.173 2.151	10 20 20 5	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	1.921 1.906 1.897 1.870	20 5b 25 20 25	-2 3 0 2 3 0 -1 -3 3 0 4 2 -1 3 3	1.633 1.620 1.607 1.594 1.586	30 70 30 10 60	-:      - 	3 1 0 3 3 0 4 3 1 -5 2 1 -3 4	
2.133 2.116 2.093 2.080	20 10 10 5	0 -2 3 2 1 1 -1 2 3 0 2 3	1.838 1.810 1.789 1.710	35 20 25 25	-2 -2 3 -2 2 3 0 0 4 2 -2 2	1.572 1.553 1.545 1.537 1.514	10 30 40 40		3 2 2 2 2 4 1 -1 4 2 0 3 3 -1 1	
2.009 1.997 1.987 1.974 1.952	35 35 20 20	-2 0 3 1 -3 2 2 -2 1 2 2 1 1 3 2	1.681 1.669 1.660 1.656	25 40 40 10 40	-1 -5 1 -2 4 0 2 4 0 0 -4 3 -3 1 2	1.505	5 90b		2 -4 3 3 -3 1	

Strong lines: 7.17/X 1.49/9 3.58/8 1.62/7 4.37/6 1.59/6 4.19/5 2.50/5

9-1488 JCPDS-ICDD Copyright (c) 1991 Quality:	ł	Carl	linte.
'9-1488       JCPDS-ICDD Copyright (c) 1991       Quality:         Al Si O (DH)       2 2 5 4         Aluminum Silicate Hydroxide         Kaolinite-1Md         Rad: CuKa       Lambda: 1.5418       Filter: Ni       d-sp:         Cutoff:       Int: Diffractometer       L/Icor:         Ref: Brindley, G., Penn State University, University Park, Pennsylvania, USA, JCPDS         Grant-in-Aid Report, (1977)         Sys: Nonoclinic       S.G.: C*/*         a: 5.16       b: 8.93       c: 7.39         A:       B: 104.5       C:         A:       B: 104.5       C:         A:       B: 104.5       C:         A:       B: 104.5       C:         Ref: Robertson, R. et al., Am. Mineral., 39 11B (1954)       Dx: 2.60         Dx:       2.60       Dm:       SS/FOM: F20=2(.112,87)         ea:       nvB:       ey:       Sign: 2V:         Ref:       Speciaen from Pugu, Tanganyika. 0=impurity, probably guartz.         Xaolinite-Serpentine group, dioctahedral subgroup. C.D. Cell: a=7.390, b=0.830, c=5.160, beta=104.50, a/b=0.8275, c/b=0.5778, S.G.=A*/*. PSC: mC34.         See original PDF Card for Graphical diffractometer trace. To replace 6-221.	d A 7.1 4.41 3.56 2.551 2.491 2.430 2.375 2.327 2.200 1.980 1.980 1.888 1.787 1.665 1.540 1.488 1.455 1.430 1.378 1.309 1.284	(a d 100 60b 100 50b 100 50 10 6 6 12 16 30 5 30 5 10 6 6 12 16 3 30 5 30 5 6 6 6 6 6 6 6 6 6 6 6 6 6	$\begin{array}{c} h \ k \ 1 \\ 0 \ 0 \ 1 \\ 1 \ 1 \ 0 \\ 0 \ 0 \ 2 \\ 1 \ 3 \ 0 \\ 2 \ 0 \ 0 \\ 2 \ 0 \ 0 \\ 0 \ 0 \ 3 \\ -1 \ 1 \ 3 \\ 2 \ 0 \ 1 \\ 2 \ 2 \ 1 \\ 0 \ 4 \ 2 \\ 0 \ 0 \ 4 \\ 1 \ 5 \ 0 \\ -2 \ 0 \ 4 \\ 1 \ 1 \ 4 \\ 0 \ 6 \ 0 \\ 3 \ 3 \ 0 \\ 0 \ 0 \ 5 \\ 0 \ 6 \ 2 \\ 2 \ 0 \ 4 \\ -4 \ 0 \ 2 \end{array}$

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Strong lines: 7.10/X 3.56/X 4.41/6 2.33/4 2.49/3 1.49/3 2.55/3 2.38/2

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Al (Si / 2 3	A1)0 (OH, 10	F) 2						9.95 /	95	20-		0 2	- 4
otassiu	e Alusinu	i Silicate Hydroxid	le					4.97 ✓	30 20	17.8 18.57	0 ( -1 1	04 11	•
uscoviti	e-2111							4.30	4		1	1 1 2 2	
ad: CuKi utoff: ef: Gil communic	a La li lery, F., ation	mbda: 1.5418 it: Diffractometer Penn State Univers	Filter [/[con sity, Univer	r: Ni r: rsity P. 	ark, Penns)	I-sp: ylvania, US	SA, Private	3.95 3.88 3.73 3.48	6 14 18 20		 -1 0 -1	1 2 1 3 2 3 1 4	
ys: Non : 5.19	oclinic	S.G.; C2 b; 9.03	2/c (15) c: 20.05		A:	C:		3.34	25		0	24	
ef: Ibi	d.	8: 95.77	C:		Zi 4	sp:			30		-0.0	0.5 1.4 1.5	
x: 2.8	4 De:	SS/FDN:	F30=12(.045	5,56)				2.987	35,		0	25	
a: 1.50	-1.56, N	B: ey: fferson An Niger	1.59-1.61	Sign (1939	:-, 2V: )	36-50 deg.	•	2.789	20.		-1	1 6	
					•••••••••			2.596	16 55/		-1 3	31	2
olor: Co pace gr	olorless oup_by_lac	kson <sub>t</sub> West, I. Kri	stallogr <sub>:1.</sub>	76 211	(1930) and	1		2.505	14		-1 (	0 8	ļ
endrick	e laffare	on. An. Hineral	24 /29 (193	59J. U	ther Source	:5		1 9 178	1		-1 -		
ive refi	ractive in	dexes for muscovit	es: alpha=1	.55-1.	57, beta=1,	.58-1.61,	E. N.	i 2.960 1 2.450	1 8 1 8		2	33 02	
ive refi amma=1. ell: a=2 15). P	ractive in 59-1.62. 20.050, b: SC: #C84.	dexes for muscovit Locality not given 9.030, c=5.190, be Volume[CD]: 934.9	es; alpha=1 n. Mica gro ta=95.77, a 10.	.55-1. oup, die 1/0=2.2	57, beta=1, octahedral 204, c/b=0.	.58-1.61, subgroup. .5748, S.G.	C.D. .=A2/a	2.450	8 10 25		-2 ( -2 ( 1 ;	3 3 0 2 0 4 3 3	
ive refi ansa=1. ell: a=1 15). PS	ractive in 59-1.62. 20.050, b: SC: mC84,	dexes for muscoult Locality not given 9.030, c=5.190, be Volume[CD]: 934.9	es; alpha=1 n. Mica gro sta=95.77, a 10.	1.55-1.3 5up, di 1/5=2.2	57, beta=1, octahedral 204, c/b=0.	.58-1.61, subgroup. .5748, S.G.	C.D. .=A2/a	2.450 2.398 2.384 2.254	8 10 25 10		-2 ( -2 ( 1 )	3 3 0 2 0 4 3 3 4 0	
ive refi anna=1. ell: a= 15). P!	s, serve in 59-1.62. 20.050, b: SC: mC84,	dexes for muscovit Locality not given 9.030, c=5.190, be Volume[CD]: 934.9	ies; alpha=1 h. Mica gro ta=95.77, a l0.	1.55-1.1 oup, die a/b=2.2	57, beta=1. octahedral 204, c/b=0.	.58-1.61, subgroup. .5748, S.G.	C.D. .=A2/a	2.450 2.450 2.398 2.384 2.254	8 10 25 10		2 -2 1	3 3 0 2 0 4 3 3 4 0	
ive refi anma=1. ell: a= 15). P 15). P	s, serie ractive in 79-1.62. 20.050, b: SC: mCB4.	dexes for muscovit Locality not given 9.030, c=5.190, be Volume[CD]: 934.9	ies; alpha=1 h. Mica gro ta=95.77, a l0.	1.55-1. 5up, di 6/b=2.2 1 Int.	57, beta=1, octahedra1 204, c/b=0.	.58-1.61, subgroup. .5748, S.G.	€.D. ,=A2/a ; d A	2.450 2.450 2.398 2.384 2.254	8 10 25 10	h k 1	2 ( -2 ( 1 )	3 3 0 2 0 4 3 3 4 0	
ive refi anma=1. ell: a= i5). P: d A .236	s, serie i ractive in 59-1.62. 20.050, b: SC: mCB4, SC: mCB4,	dexes for muscovit Locality not given 9.030, c=5.190, be Volume[CD]: 934.9 h k l -1 3 5 2 2 1	ies; alpha=1 h. Mica gro ta=95.77, a h0. i d A i 1.646 i 1.631	.55-1. pup, dia /b=2.2 1 [nt. 1 25 5	57, beta=1, octahedra1 204, c/b=0.	.58-1.61, subgroup. .5748, S.G. k l 3 9 5 4	C.D. ,=A2/a ; d A ; 1.267 ; 1.253	2.450 2.398 2.398 2.384 2.254 ( Int. (	8 8 10 25 10	h k l 4 13 2 14	2 ( -2 ( 1 )	3 3 9 2 0 4 3 3 4 0	
ive refi anna=1. ell: a= 15). P: d A .236 .208 .189 .149	s, serie ractive in 59-1.62. 20.050, b: SC: mCB4, SC: mCB4, Int.1 4 4 4 4 4	dexes for muscovit Locality not given 9.030, c=5.190, be Volume[CD]: 934.9 h k l -1 3 5 2 2 1 0 2 8 2 2 2	tes; alpha=1 h. Mica gro ta=95.77, a h0. i d A i 1.646 i 1.631 i 1.620 i 1.603	.55-1. pup, dia /b=2.2 1 Int. 1 25 6 6 6 6 6	57, beta=1, octahedra1 204, c/b=0.	.58-1.61, subgroup. .5748, S.G. . k l	C.D. ,=A2/a ; d A ; 1.267 ; 1.253 ; 1.246 ; 1.227	2.450 2.398 2.398 2.384 2.254 ( Int. ( 4 6 8 4	8 8 10 25 10	h k l 4 13 2 14 5 0 5 2	2 0	3 3 3 0 2 0 4 3 3 4 0	
ive refi anma=1. ell: a= 15). P 15). P d A .236 .208 .189 .149 .132	s, serie i ractive in 20,050, b: SC: mCB4, SC: mCB4, Int. 1 4 8 4 16 20	dexes for muscovit Locality not given 9.030, c=5.190, be Volume[CD]: 934.9 h k l -1 3 5 2 2 1 0 2 8 2 2 2 1 3 5	tes: alpha=1 h. Mica gro ta=95.77, a h. h. h. h. h. h. h. h. h. h. h. h. h.	55-1. 500, dia 100, dia	57, beta=1, octahedra1 204, c/b=0. 1 1 1 1 2 -2 -2 -3	.58-1.61, subgroup. .5748, S.G. .5748, S.G. .5748, S.G. .5748, S.G. .54 .4 .5 .4 .5 .4 .5 .4 .5 .1 .6	C.D. =A2/a { d A { 1.267 { 1.253 { 1.246 { 1.227 { 1.221}} {	2.450 2.398 2.398 2.384 2.254 ( Int. ( 4 6 8 4 6	8 8 10 25 10	h k l 4 13 2 14 3 5 0 5 2 7 4	2 -2 1 0		
ive refi anna=1. ell: a= 15). P: 15). P: 15). P: 15). 206 .208 .189 .149 .132 .070 .053	s, serie i ractive in 20,050, b: SC: mCB4, SC: mCB4, Int. 10, 10 16 20 4 16 20 4	dexes for muscouit Locality not given 9.030, c=5.190, be Volume[CD]: 934.9 h k l -1 3 5 2 2 1 0 2 8 2 2 2 1 3 5 2 2 3 0 4 4	es; alpha=1 h. Mica gro ta=95.77, a h. i d A i 1.646 i 1.631 i .620 i 1.603 i 1.573 i .573 i 1.559 i 1.541	55-1. 50, dia 10, b=2.2 11, lot 12, lot 10,	57, beta=1, octahedra1 204, c/b=0. 1 1 1 1 2 -2 -2 -3 -2 -3 -2 -2 -3	58-1.61, subgroup. 5748, S.G. 5748, S.G. 4 5 5 4 4 3 4 5 1 6 2 10 5 6	C.D. =A2/a d A 1.267 1.253 1.246 1.227 1.221 1.208 1.200 1.200	2.450 2.398 2.398 2.384 2.254 ( Int. ( 4 6 8 4 6 8 4 4 6		h k l 4 13 2 14 3 5 0 5 2 7 4 5 3 2 3	2 0	3 3 0 2 0 4 3 3 4 0	
ive refi anna=1. ell: a= 15). P: d A .236 .208 .149 .132 .070 .053 .993 .972	s, serie i ractive in 20,050, b: SC: mCB4, SC: mCB4, Int. 10, 4 4 4 4 4 5, 10, 10, 10, 10, 10, 10, 10, 10, 10, 10	dexes for muscoult Locality not given 9.030, c=5.190, be Volume[CD]: 934.9 h k l -1 3 5 2 2 1 0 2 8 2 2 2 1 3 5 2 2 3 0 4 4 0 0 10 -1 3 7 -2 0 6	tes: alpha=1 Mica gro ta=95.77, a 10. 1.646 1.646 1.631 1.620 1.603 1.524 1.524 1.504 1.504	55-1. 500, dia 100, dia	57, beta=1, octahedral 204, c/b=0. 1 1 -1 -2 -2 -3 -2 -1 () -2 -1 -1	58-1.61, subgroup. 5748, S.G. 5748, S.G. 4 5 5 4 4 3 4 5 1 6 2 10 5 6 3 11 4 7 2 13	C.D. =A2/a d A 1.267 1.253 1.246 1.227 1.221 1.208 1.200 1.1828 1.1828	2.450 2.398 2.398 2.384 2.254 2.254 (Int.) 4 6 8 4 6 8 4 4 6 4 4 4 4 4 4 4 4 2		h k l 4 13 2 14 3 5 0 5 2 7 4 3 5 3 2 15 7 6 9 6 1	2 ( -2 ( 1 0		
ive refi amma=1.: ell: a= 15). P: d A .236 .208 .189 .149 .132 .053 .993 .972 .951 .941	s, serve in ractive in 29-1.62. 20.050, b: SC: mCB4, SC: mCB4, Int. 10 4 4 4 4 4 4 5 4 5 4 5 4 5 4 5 4 5 4 5	dexes for muscovit Locality not given 9.030, c=5.190, be Volume[CD]: 934.9 h k l -1 3 5 2 2 1 0 2 8 2 2 2 i 3 5 2 2 3 0 4 4 0 0 10 -1 3 7 2 0 6 -2 2 6	tes: alpha=1 Mica gro ta=95.77, a 10. 1.646 1.631 1.620 1.603 1.573 1.559 1.541 1.524 1.524 1.504 1.453 1.424	55-1. 50, dia 10, b=2.2 11, lot 12, lot 14, lot 12, lot 14,	57, beta=1, octahedral 204, c/b=0. 1 1 -1 -2 -2 -3 -2 -3 -2 -1 () -2 0	58-1.61, subgroup. 5748, S.G. 5748, S.G. 4 5 4 4 3 4 5 1 6 2 10 5 6 3 11 4 7 2 13 0 14	C.D. =A2/a d A 1.267 1.253 1.246 1.227 1.221 1.208 1.200 1.1903 1.1828 1.1582 1.1300	2.450 2.398 2.398 2.398 2.398 2.398 2.254 2.254 (Int.) 4 6 8 4 4 6 8 4 4 4 4 4 4 4 2 2		h k l 4 13 5 0 5 2 7 4 5 3 2 15 7 4 5 3 2 15 7 6 11 7 6 11 8 6 8	2 0	3 3 3 0 2 3 3 4 0	
ive refi anna=1.: ell: a= 15). P: 15). P: 15].	s, serie in ractive in 59-1.62. 20.050, b: SC: mCB4, SC: mCB4, Int. 10 4 4 4 4 4 4 5 4 5 4 5 4 4 4 6 4 4 5 4 2 4 4 4 5 4 2 4	dexes for muscovit Locality not given 9.030, c=5.190, be Volume[CD]: 934.9 h k l -1 3 5 2 2 1 0 2 8 2 2 2 i 3 5 2 2 3 0 4 4 0 0 10 -1 3 7 2 0 6 -2 0 8 1 3 7	tes: alpha=1 Mica gro ta=95.77, a 10. 1.646 1.631 1.620 1.603 1.573 1.559 1.541 1.524 1.524 1.504 1.453 1.424 1.414 1.388	55-1. 50, dia 50, d	57, beta=1, octahedral 204, c/b=0. 1 1 -1 2 -2 -3 -3 -2 -1 () -2 0 0 0	58-1.61, subgroup. 5748, S.G. 5748, S.G. 1 k l 3 9 5 4 4 3 4 5 1 6 2 10 5 6 3 11 4 7 2 13 0 14 4 11 5 8	C.D. =A2/a d A i.267 i.253 i.246 i.227 i.221 i.208 i.200 i.1903 i.1828 i.1582 i.1300 i.1220 i.1167	2.450 2.398 2.398 2.398 2.398 2.254 2.254 1.11 4 6 8 4 4 6 8 4 4 4 4 4 4 4 4 4 4 4 4 4		h k l 4 13 2 14 5 0 5 2 7 4 5 3 2 15 6 11 6 12 6 12	2 ( -2 ( 1 0		
ive refi amma=1. ell: a= 15). P d A .236 .208 .189 .149 .132 .070 .053 .993 .972 .993 .972 .951 .894 .871 .894 .871 .822 .746	s, serie in ractive in 59-1.62. 20.050, b: SC: mC84, Int. 4 4 4 16 20 4 5 5 10 6 4 4 4 4 4 4 4 4 4 4 4 4 4	dexes for muscovit Locality not given 9.030, c=5.190, be Volume[CD]: 934.9 h k l -1 3 5 2 2 1 0 2 8 2 2 2 1 3 5 2 2 3 0 4 4 0 0 10 -1 3 7 2 0 6 -2 2 6 -2 0 8 1 3 7 0 2 10 -2 2 8	tes: alpha=1 Mica gro ta=95.77, a 10. 1.646 1.631 1.620 1.603 1.573 1.559 1.541 1.524 1.524 1.524 1.504 1.423 1.424 1.424 1.388 1.375 1.352	55-1. 50, dia 50, d	57, beta=1, octahedra1 204, c/b=0. 1 1 -1 2 -2 -3 -2 -1 -1 -1 (clu) -2 0 0 0 0 0 1 -3 -1	58-1.61, subgroup. 5748, S.G. 5748, S.G. 1 k l 3 9 5 4 4 3 4 5 1 6 2 10 5 6 3 11 4 7 2 13 0 14 4 11 5 8 3 7 3 13	C.D. =A2/a d A i.267 i.253 i.246 i.227 i.221 i.208 i.200 i.1903 i.1828 i.1582 i.1300 i.1220 i.1220 i.1167	2.450 2.398 2.254 2.254 2.40 2.398 2.398 2.398 2.398 2.254 2.398 2.398 2.398 2.254 2.254 2.398 2.398 2.40 2.40 2.40 2.40 2.40 2.40 2.40 2.40		h k l 4 13 2 14 3 5 0 5 2 7 4 3 5 3 2 3 2 15 7 6 6 11 2 6 8 5 13 6 12	2 0		
ive refi amma=1. ell: a= 15). P d A .236 .208 .189 .149 .132 .070 .053 .993 .972 .951 .971 .894 .871 .822 .746 .731	s, serie i ractive in 59-1.62. 20.050, b: SC: mC84, SC: mC84, 10 10 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4	dexes for muscovit Locality not given 9.030, c=5.190, be Volume[CD]: 934.9 h k l -1 3 5 2 2 1 0 2 8 2 2 2 1 3 5 2 2 3 0 4 4 0 0 10 -1 3 7 2 0 6 -2 2 6 -2 0 8 1 3 7 0 2 10 -2 2 8 -1 3 9	tes: alpha=1 Mica gro ta=95.77, a 10. 1.646 1.646 1.631 1.620 1.603 1.573 1.559 1.559 1.541 1.524 1.504 1.504 1.424 1.414 1.388 1.375 1.352 1.352 1.355	55-1. 50, dia 50, d	57, beta=1, octahedra1 204, c/b=0. 1 1 -1 2 -2 -3 -1 -1 -1 -1 -1 -1 -1 -2 -3 -3 -2 -1 -1 -1 -1 -1 -2 -3 -3 -2 -1 -1 -1 -1 -2 -3 -3 -2 -1 -1 -1 -2 -3 -3 -3 -3 -3 -3 -3 -3 -3 -3 -3 -3 -3	58-1.61, subgroup. 5748, S.G. k l 3 9 5 4 4 3 4 5 1 6 2 10 5 6 3 11 4 7 2 13 0 14 4 11 5 8 3 7 3 13 3 8	C.D. =A2/a d A 1.267 1.253 1.246 1.227 1.221 1.208 1.200 1.1903 1.1828 1.1582 1.1300 1.1220 1.1167	2.465 2.450 2.398 2.384 2.254 2.254 1 Int.1 4 6 8 4 4 6 4 4 4 4 4 4 4 4 4 4 4 4 4 4		h k l 2 14 3 5 0 5 2 7 4 3 5 3 2 15 7 6 6 11 6 12 6 12			
ive refi anna=1.: ell: a= i5). P: d A .236 .200 .189 .149 .132 .070 .053 .993 .972 .973 .971 .8941 .8941 .822 .746 .731 .710 .704	s, gerein ractive in ractive in 20.050, b: SC: mCB4, SC: mCB4, Int. 4 4 4 5 4 4 5 4 4 5 4 4 5 4 4 4 4 4 4	dexes for muscovit Locality not given 9.030, c=5.190, be Volume[CD]: 934.9 h k l -1 3 5 2 2 1 0 2 8 2 2 2 1 3 5 2 2 3 0 4 4 0 0 10 -1 3 7 2 0 6 -2 2 6 -2 0 8 1 3 7 0 2 10 -2 2 8 -1 3 9 2 0 8 -1 5 1 -5 1	tes: alpha=1 Mica gro ta=95.77, a 10. 1.646 1.646 1.631 1.620 1.603 1.573 1.559 1.541 1.524 1.504 1.504 1.424 1.414 1.388 1.424 1.414 1.388 1.375 1.299 1.299 1.597 1.559 1.559 1.559 1.559 1.559 1.352 1.559 1.559 1.559 1.352 1.559 1.	55-1. 50, dia 50, d	57, beta=1, octahedra1 204, c/b=0. 1 -1 -1 -2 -3 -2 -3 -2 -1 -1 -1 -1 -1 -1 -1 -1 -1 -1 -1 -1 -1	58-1.61, subgroup. 5748, S.G. 5748, S.G. k l 3 9 5 4 4 3 4 5 1 6 2 10 5 6 3 11 4 7 2 13 0 14 4 11 5 8 3 7 3 13 3 8 2 13 6 9 1 4 9 5 4 4 3 4 5 1 6 3 11 4 7 2 13 0 14 4 1 5 8 3 7 3 13 9 9 9 1 4 1 5 8 3 7 3 13 9 9 1 4 1 5 1 6 1 6 1 6 1 6 1 7 1 6 1 7 1 6 1 7 1 7 1 7 1 7 1 7 1 7 1 7 1 7	C.D. ,=A2/a d A 1.267 1.253 1.246 1.227 1.221 1.208 1.200 1.1903 1.1828 1.1582 1.1300 1.1220 1.1167	2.465 2.450 2.398 2.398 2.398 2.254 2.254 1 Int.1 4 6 8 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4		h k l 4 13 2 14 5 0 5 2 7 4 5 3 2 15 7 6 6 11 6 8 5 13 6 12	-2 -2 -2 -2 -2 -2 -2 -2 -2 -2 -2 -2 -2 -	3 3 3 3 9 2 3 3 9 4 3 3 9 4 3 3 9 4 4 4 4 4 4 4 4 4	
ive refi anna=1.: ell: a= 15). P: d A .236 .208 .149 .132 .208 .149 .132 .993 .993 .972 .9351 .941 .894 .871 .822 .746 .731 .710 .704 .699 .662	s, serie in ractive in 59-1.62. 20.050, b: SC: mCB4, Int.1 4 16 20 4 4 16 20 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4	dexes for muscovit Locality not given 9.030, c=5.190, be Volume[CD]: 934.9 k k l -1 3 5 2 2 1 0 2 8 2 2 2 1 3 5 2 2 3 0 4 4 0 0 10 -1 3 7 2 0 6 -2 2 6 -2 0 8 1 3 7 0 2 10 -2 2 8 -1 3 9 2 0 8 -1 3 9 2 0 8 -1 5 1 -3 1 1 0 0 12	tes: alpha=1 Mica gro ta=95.77, a 10. 1.646 1.646 1.631 1.620 1.603 1.573 1.559 1.541 1.524 1.525 1.521 1.229 1.292 1.274	55-1. 50p, dia 50 - 2.2 10 - 4 12 10 - 4 10 - 4	57, beta=1, octahedra1 204, c/b=0. 1 1 -1 2 -2 -3 -1 -1 -1 2 -2 -3 -1 -1 -1 -1 -1 -1 -1 -1 -2 -3 -3 -2 -3 -1 -1 -1 -1 -2 -3 -3 -2 -3 -2 -2 -3 -2 -2 -3 -2 -2 -3 -2 -2 -3 -2 -2 -3 -2 -2 -3 -2 -2 -3 -2 -2 -3 -2 -2 -3 -2 -2 -3 -3 -2 -2 -3 -3 -2 -2 -3 -3 -2 -1 -1 -1 -1 -1 -2 -2 -3 -3 -2 -1 -1 -1 -1 -1 -2 -2 -3 -3 -1 -1 -1 -1 -1 -1 -1 -1 -1 -1 -1 -1 -1	58-1.61, subgroup. 5748, S.G. k l 3 9 5 4 4 3 4 5 1 6 2 10 5 6 3 11 4 7 2 13 0 14 4 11 5 8 3 7 3 13 3 8 2 13 6 0 3 9 6 4	C.D. ,=A2/a d A i.253 i.246 i.227 i.221 i.200 i.1903 i.1828 i.1582 i.1300 i.1220 i.1167	2.465 2.450 2.398 2.384 2.254 2.254 1 Int.1 4 6 8 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4		h k l 4 13 2 14 5 2 7 4 5 3 2 15 7 4 5 13 6 12 6 12	2 ( -2 ( 1 )	3 3 3 2 4 3 3 0 4 3 3 0 4 3 4 0	

7-25 JCPDS-ICDD Copyright (c) 1991 Quality: i		Musc.
7-25       JCPDS-ICDD Copyright (c) 1991       Quality: i         KA1 Si A10 (OH)       2       3       10       2         Potassium Aluminum Silicate Hydroxide       Huscorite-IM, syn       d-sp:         Rad: CuKa       Lambda: 1.5418       Filter: Ni       d-sp:         Cutoff:       Int: Diffractometer       1/Icor:       Ref: Yoder, H., Eugster, H., Geochia. Cosmochia. Acta, 8 225 (1955)         Sys: Monoclinic       S.6.: C2/m (12)       A:       C:         A:       B: 101.6       C:       I:       2         A:       B: 101.6       C:       I:       2       mp:         Ref: Ibid.       Dx:       2.81       Dm:       2.83       SS/FOM: F29=9(.037,84)         ea: 1.563(11), nvB: 1.596(14), ey: 1.602(15), Sign: -, 2V: 30-47 deg.       Ref: Ibid.         Color: Cohorless. or light shades of green-red or brown       Color:       Cohorless. or light shades of green-red or brown	d A 10.1 5.04 4.49 4.35 4.11 3.66 3.36 3.07 2.929 2.689 2.582 2.565 2.550 2.450 2.405 2.246 2.219 2.191	Multer . Int.: h k l 100 0 0 1 35 0 0 2 90 0 2 0 25 -1 1 1 16 0 2 1 60 -1 1 2 100 0 0 3 50 1 1 2 6 -1 1 3 16 0 2 3 50 -1 3 0 90 -1 3 1 20 2 0 0 12 1 3 1 4 -1 3 2 12 -1 1 4 B 0 4 0 B 2 2 0 4 0 4 1
CAS no.: 1318-94-1. Synthesized from KAISiD4 + kaolinite + water at 200 C and 15,000 P.S.I. for 4120 hours. Mica group, dioctahedral subgroup. C.D. Cell: a=10.275, b=8.995, c=5.208, beta=101.60, a/b=1.1423, c/b=0.5790, S.B.=A2/m (12). PSC: mC42. To replace 2-56. Mwt: 398.31. Volume(CD1: 471.51.	2.156 2.109 2.013 1.957 1.900 1.668	20 -1 3 3 6 2 0 2 30 0 5 8 1 3 3 4 -1 3 4 18 -2 4 2
dA ; Int.; hkl i dA i Int.; hkl i dA	Int.	h k 1
1.653     12     151       1.635     12     204       1.514     4     135       1.499     35     060		

Strong lines: 10.1/X 3.36/X 4.49/9 2.57/9 3.66/6 3.07/5 2.58/5 5.04/4

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;- 710 JCPDS-ICDD Copyright (c) 1991 Quality: i		Pu	rite
٤،٩	d A	Int.	h k 1
2 Iron Sulfide	3.128 2.709 2.423	35 85 65	1 1 1 2 0 0 2 1 0
Pyrite, syn	2.2118	50 40	2 1 1 2 2 0
Rad: CuKal Lambda: 1.5405 Filter: Ni d-sp: Cutoff: Int: Diffractometer I/Icor: Ref: Svanson et al., Natl. Bur. Stand. (U.S.), Circ. 539, 5 29 (1955)	1.6332 1.5640 1.5025	100 14 20	3 1 1 2 2 2 2 3 0 3 2 1
Sys: Cubic S.G.; Pa3 (205)	1.2427	12	3 3 i
a: D.417 D: L: N: C. A: B: C: Z: 4 sp: 642 C Ref: Ibid.	1.2113 1.1823 1.1548	14 8 6	4 2 0 4 2 1 3 3 2
Dx: 5.01 Da: 5.02 SS/FDN: F24=22(.029,37)	1.1057	6     25	4 2 2 5 1 1
ea: nwB: ey: Sign: 2V: Ref:	1.0060 0.9892 0.9577	8 6 12	2 5 0 5 2 1 4 4 0
Color: Black (in powder), brass-yellow (in crystals) X-ray pattern at 26 C. CAS no.: 1309-36-0. Sample prepared as a fine precipitate and beated in a closed tube in S2 atmosphere for 4 hours at 700 C.	0.9030 0.8788	16 1 8	600 611
Spectroscopic analysis: (0.11 AI, Ca, Mg, Si; (0.011 Co, Cu, Mo, Ni, Pb; (0.0011 Cr, Ge, Mn; (0.00011 Ag, Validated by calculated pattern 24-76.	0.8565 0.8261 0.8166	8 1 4 1 4 1	620 533 622
RP2Re=51.7, Disp.=16, VHN100≈1505-1620, Color values=.327, .335, 51.8, Ref.: IMA Commission on Ore Microscopy ODF. Measured density and melting point by Dana's System of Mineralogy, 7th Ed., 1 238. FeS2 type. Pyrite group, pyrite subgroup. Also called: pyrites. Also called: fools gold. PSC: cP12. To be deleted by Z-506, lover Fn, Bayliss, 11/90. Mwt: 119.97.	0.7981	6	6 3 Ī

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Strong lines: 1.63/X 2.71/9 2.42/7 2.21/5 1.92/4 3.13/4 1.44/3 1.04/3

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33-1161	JCPDS-1CDD Copyright (c) 1991	Quality: +	6	tuants
: SiO			d A   Int.	h k 1 - )**
2 Jilicon Oxide Quartz, syn		26 26.7°	4.257 22 3.342 100 2.457 8 2.282 8	
`ad: CuKal Cutoff: Ref: Natl. Bur.	Lambda: 1.540598 Filter: Mon- Int: Diffractometer [/Icor: 3.6 Stand. (U.S.) Monogr. 25, 18 61 (198	o. d-sp: Diff. 1)	2.127 6 <sup>-</sup> 1.9792 4 1.8179 14 <sup>-</sup>	
Sys: Hexagonal a: 4.9133(2) A: zf: Ibid.	S.G.: P3221 (154) b: c: 5.4053(4) B: C:	A: C: 1.1001 Z: 3 mp:	1.6719 4 1.6591 2 1.6082 (1 1.5418 9	2 0 2 1 0 3 2 1 0 2 1 1
3: Ref: Svanson, Fu	nvB: 1.544, ey: 1.553, Sig yat, Natl. Bur. Stand. (U.S.), Circ.	n: + 2V; 539, 3 24 (1954)	1.3820 6 1.3752 7 1.3718 8	$\begin{array}{c} 1 & 1 & 3 \\ 3 & 0 & 0 \\ 2 & 1 & 2 \\ 2 & 0 & 3 \\ 3 & 0 & 1 \end{array}$
Jor: Colorless Pattern at 25 C. USA, ground sing and G. McCarthy, .PDS Grant-in-A calculated patter used as internal 0.9089. Mut: 60	Sample from the Glass Section at Ni le-crystals of optical quality. Path North Dakota State University, Fargo id Report (1990). Agrees well with e rns. O2Si type. Quartz group. Also standard. PSC: hP9. To replace S-4 .00. Volume[CD]: 113.00.	BS, Gaithersburg, Maryland, tern reviewed by J. Holzer o, North Dakota, USA, experimental and o called: silica. Silicon 490. Plus 6 reflections to	1.2880       2         1.2558       2         1.2558       2         1.1999       2         1.1978       1         1.1843       3         1.1804       3	2 2 0 2 1 3 2 2 1 1 4 3 1 0
d A   Int. 1532   1	h k l ; d A ; int. 3 1 1 ; 1.0476 ; 1	1 0 5 0.9873		
1.1405 (1 1.1143 (1 1.0913 2 0635 (1	2 0 4 1.0436 (1 3 0 3 1.0347 (1 3 1 2 1.0150 1 4 0 0 0.9898 1	2 1 4 0.9762 2 2 3 0.9636 4 0 2	(1	3 2 0 2 0 5

trong lines: 3.34/X 4.26/2 1.82/1 1.54/1 2.46/1 2.28/1 1.37/1 1.38/1

-696	JCPDS-ICDD Copyright (c)	1991	Quality: #			Sid	Cent	ق
FeCO 3 Iron Carbonate Siderite Rad: CuKal Cutoff: Ref: Natl. Bur.	Lambda: 1.540598 Int: Diffractometer Stand. (U.S.) Monogr. 2	Filter: Mono. 1/lcor: 5, 15 32 (1978	, d-:	39° 2	·	d A 3.593 2.795 2.564 2.346 2.134 1.965 1.7968 1.7368 1.7315	Int.: 25 100 (1 20 20 20 12 30 35	h k 1 0 1 2 1 0 4 0 0 6 1 1 0 1 1 3 2 0 2 0 2 4 0 1 8 1 1 5
Sys: Rhombohed a: 4.6935(2) A: Ref: lbid. Dx: 3.93 ea: Ref: Dana's S Color: Light Pattern at 25 Spectrographi Camborne Ca PSC: hRIO.	ral (Hex) S.G.: R-3c b: C: B: C: Dm: 3.B9 SS/FDM: F3/ nwB: 1.8728(1), ey: 1 ystem of Mineralogy, 7th yellowish brown C. Specimen from Lwigte c analysis indicates 1-2 hcite group, calcite subg b replace 8-133. Mwt: IT	(167) 15.386(8) 0=75(.010,39) .6331, Sig Ed., 2 167 it, Greenland. I Mn. Bptical roup. Silicop 5.86. Volume	A: 2: 6 (n: - 2V: (NNNH 1328 data species used as jot CDJ: 293.53	C: 3.2782 mp: ////////////////////////////////////	ard.	1.5291 1.5063 1.4266 1.3969 1.3818 1.3818 1.3283 1.2823 1.2209 1.2002 1.2002 1.1977 1.1254 1.1154 1.0872	3 1000 10-000 AMA -0	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$
d A   Int 1.0820   5 1.0671   4 0.9825   5 0.9724   5 0.9724   5	h k l 1 3 4 2 2 6 4 0 4 3 1 8 2 0 14	d A   Int 0.9358   2 0.9309   6 0.9256   3	1 1 3 2	k 1 0 16 2 1 3 2	d A	Int.		h k l

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Strong lines: 2.80/X 1.73/4 1.74/3 3.59/3 2.35/2 2.13/2 1.97/2 1.51/1

3-558 JCPDS-ICDD Copyright (c) 1991 Buality: i		Talc
3-558       JCPDS-ICDD Copyright (c) 1991       Buality: i         Ng Si O (OH) 3 4 10 2	d A 9.34 4.66 4.55 3.51 3.43 3.116 2.892 2.595 2.476 2.335 2.212 2.196 2.122 2.103 1.930 1.870 1.557 1.557 1.557	alc $1nt.$ h k 1 $1004 = 0$ 0 $908 = 0$ 0 $30$ -i $1$ 1 $4$ -1 $1$ 1 $1$ 1 $1$ 1 $1$ 1 $1$ 0 $1$ 0 $1$ 0 $1$ 0 $1$ 1 $1$ 1 $1$ 1 $1$ 1 $1$ 1 $1$ 1 $1$ 1 $1$ 0 $1$ 0 $2$ $2$ $30$ -1 $30$ -1 $1$ $3$ $15$ $2$ $1$ $1$ $20$ $2$ $20$ $2$ $20$ $2$ $20$ $1$ $20$ $1$ $2$ $2$
b = 3, 156, C = 3, 27, 0 = 4 = 5, 53, 27, 0 = 4, 0 = 0, 57, 3, 3, 4, -7, 74, (13), 4 = 1, 14, 14, 14, 15, 16, 16, 16, 16, 16, 16, 16, 16, 16, 16	1.509 1.460 1.406 1.394	10 3 3 0 8bi 3 3 2 16 3 1 6 20 -1 3 12

,

Strong lines: 9.34/X 3.12/X 4.66/9 2.48/7 1.87/4 1.53/4 4.55/3 2.60/3

# APPENDIX C

# METALLURGICAL TEST RESULTS

D.D.H. Geomanagement Ltd. 422-470 Granville St. Vancouver, B.C. V6C 1V5 December 23, 1992

Attention: A.D. Drummond

Subject: Laboratory Evaluation / Quinto Gold / Graphite Deposit.

This will confirm the discussion which took place in a meeting in Quinto's office on December 21, 1992 in which the processing options for this material were reviewed.

The Quinto deposit contains both gold ( est 0.18 - 0.20 oz/t ) and graphite, as well as pyrite. The graphite and the pyrite grades have not been determined, but it is thought that the "mineable" portion of the deposit may contain several percent of each.

This suggests the possible recovery of both gold and graphite into saleable products.

Previous testing ( Lakefield and Coastech ) indicate that gold can be recovered, by cyanidation providing the carbon-in-leach procedure is used to counter the modest preg-robbing character of the graphite. CLL leaching, after grinding, appears to yield gold recoveries of + 95 %, and about 88 % using conventional cyanidation.

Heap leaching was evaluated, on samples crushed to minus 6 mesh, yielding gold recoveries of about 50 %. Based upon a comparison with the tests which were performed on ground samples, the loss in recovery would appear to be a function of "grind" rather than preg-robbing carbon. This is consistent with the optical microscopy photographs which were viewed, and which indicated that the gold is quite fine ( a few microns ) and is predominantly located along natural "fracture" lines in the pyrite.

Flotation of unweathered ore produced good gold recoveries, at 98%, but the low [4:1] ratio of concentration in rougher flotation resulted in a low concentrate grade. It was suggested that these results, when compared to CIL cyanidation, were not encouraging, and no further studies were recommended.

Gravity concentration ( Lakefield test No.3 / 1987 ) indicated that 28 % of the gold could be recovered into a gravity concentrate grading • 60 oz/t Au. Because of the small quantity of sample which was used in this test, it was not possible to evaluate additional upgrading of the first cleaner gravity concentrate. Mineral Processing Potential

The previous studies have indicated that cyanidation will effectively recovery the gold, providing CLL leaching is used.

What needs to be determined is whether an effective separation can be made between the graphite and pyrite to produce a graphite concentrate which is sufficiently depleted in pyrite to make it saleable.

It is hoped that grinding will differentially liberate the graphite and pyrite so that the former can be recovered as a "clean" flotation concentrate, while leaving the pyrite for subsequent flotation concentration of the contained gold. If this is successful, the gold / pyrite flotation concentrate could be leached, or sold if the gold grade is sufficiently high.

Since some portion of the gold is "liberated" during comminution, and since gold is readily floated, it may occur that a significant portion ( the data suggests +20% ) of the gold may be recovered with the graphite, which is not be desirable.

If this occurs, other process flowsheets need to be considered, as follows:

- (1) Leach the entire ore after grinding, then perform differential flotation on the leached tailing.
- (2) Leach the graphile flotation concentrate for gold recovery, then dewater and sell the graphile concentrate.

Testing Proposal

The following testing program was agreed upon's

Three samples, all apparently "fresh" and all collected from the underground exploration mining program, were provided for testing, as follows:

- (1) About 20 kg of ~20 mesh graphitic sample, which graded about 0.03 oz/t Au.
- (2) A single rock weighing about 2 kg and having a high pyrite content thought to grade 1 oz/t Au.
- (3) About 3 kg of lower grade sample.

These will be assayed separately for gold, and will be composited to produce a sample grading 0.18 - 0.20 oz/t Au. Undoubtedly, this sample will contain both graphite and pyrite. Ŷ

The composite will be ground to an estimated 50 % - 200 mesh, and will be subjected to gravity concentration followed by staged flotation to evaluate the natural differential flotation characteristics of the gold, graphite, and pyrite.

Only once this test has been evaluated, will subsequent testing be performed.

Cost Estimate

It is estimated that no more than 3 tests will be required to determine the potential for the production of a graphite concentrate.

The cost of this program will not exceed \$ 1,000, including assaying, although an additional allowance of \$ 500 should be provided for possible optical microscopy.

If the testing is encouraging at that time, an additional \$ 2,000 should be allocated to further investigate process parameters, such as optimum grind, and flotation reagent requirements.

Comments

To the best of my knowledge, there is no industrial precedent for this approach, so the study cannot really upon published literature. Nevertheless, the potential for success should be readily observed within the proposed 3 initial tests.

Thank you

G.Hawthorn, P.Eng

(g-0305)

DDI	4	- 6	67-	316	/				
						W- 9	2078		
GRAVITY	/ +	LOTATION	TEST: 0	iF - 1		Jan 3	3, 199	93	
CLIENT:	:	DDH Geoma	nagemer	)t	PROJEC	CT: Qu	into	/ Lum	юy
SAMPLE:	:	C+1 ( 73 ' grab samp ).Composi	% DDH s le from te grac	ample ( underg le appro	0.24 0.24 ground pximate	oz∕t ero.e ely 0.	Au + 30 oz/ 2 oz/	27 % t Au t Au.	Au-pyrite
		Note that to -20 me	the DD sh.	H samp	le had	atrea	ady be	en dr	y ground
OBJECTI	ECTIVE: Evaluate gravity concentration followed by flotation in an attempt to separate the graphite and the gold-pyrite.						lotation he		
PROCEDU	JRE :	Grind: 1,6 Pan: K = - Float: Sta Wet/dry se	000 gm + 200:1 aged ro creen r	/ 5 mìr Ngher 1 Ougher	o / 67 to comp tailin	% sol bletio ng to	ids m. 325 m	esh.	
TEST CO	ND I 1	IONS: FID	tation						
Time E	vent	DF 250	PAX	3418A	an an 17				рН
0 F 3 8 13 17 22 23 26 F	- 1	125 75 75 50 50 50	25				DF	RAF	-T
27 31	£	25 50	25						8.5
33 F	inie	th 625	50					*****	
Noter		(1) ALL re	agent	additic	ns in	g∕t o	fori	ginal	feed.

### METALLURGICAL CALCULATIONS:

Assav Distribution % Au oz/t LOI % Fe % S % Product Wt % Au \_ \_ \_ \_ \_ -----\_ \_ \_ \_ \_ \_ 1.1 2.28 Pan conc 9.1 15.2 F-1 RC 
 15.2
 0.136
 11.3
 5.5
 2.3
 7.6

 20.9
 0.980
 25.1
 34.4
 35.6
 74.6
 0.136 11.3 5.5 2.3 7.6 F-2 RC \*\*\*\*\*\*\*\* F RC 36.1 82.2 62.8 0.038 Tails 8.7 Feed (calc) 100.0 ( 0.275 ) 100.0 0.18 8888Y CC - cleaner concentrate CT - cleaner tailing Noter RC - rougher concentrate RT - rougher tailing SCREEN ANALYSIS: Flotation rougher tailing W1 % Au oz/t Mesh - - - - - - ----------22.8 0.007 DRAFT 150 13.2 0.032 200 16.6 0.020 325 47.4 0.060 ??? . . . . . . . . 100.0 ( 0.038 )

64.0 % - 200 mesh.

### NOTES:

- -- For purposes of economy in assaying, the graphite content of the first flotation concentrate, is. the graphite concentrate, was approximated using a loss-on-ignition (LOI ) analysis, rather than the more expensive carbon analysis. For purposes of comparison, a LOI analysis was performed on the second flotation concentrate, is. pyrite concentrate as well.
- -- The gold grade of the 325 mesh tailing fraction will need to be evaluation in future testing, since possible contamination from a prior test may have occurred.

#### OBSERVATIONS:

- There was no optically visible gold in the pan concentrate. The product consisted almost entirely of pyrite.
- There were virtually no naturally floatable minerals in the flotation feed, although the "graphite" floated well, but slowly, on the addition of Dowfroth 250.
- There was "no" optically discernable pyrite in the F-1 concentrate.
- The F-2 concentrate floated almost immediately upon the addition of PAX, and went to "completion" within about 6 minutes.

This product appeared to contain minor "graphite".

### CONCLUSIONS:

DRAFT

- -- See comments under NOTES
- -- The recovery of gold, at 91.3 % justifies further investigation. Although the majority of the gold reported to the F-2, ie. pyrite, flotation concentrate, the grade of this product, at 0.98 oz/t Au, is too low to justify direct marketing. Previous testing, by others, however indicates that the gold is amenable to cyanidation using the CLL procedure.
- -- The gold grade of the "graphite" concentrate is relatively low, but probably is sufficient to justify cyanidation.
- -- The F-1 "graphite" concentrate is perplexing in that while it appeared to consist almost entirely of graphite, the chemical analysis suggests otherwise. The material contains pyrite ( the Fe:S ratio indicates a possible second iron mineral), but the loss-on-ignition (LOI) is very low, and not consistent with the impression that the black mineral is graphite. This requires further investigation which will include additional assaying and optical microscopy.
- -- The pan concentrate contained only 9.1 % of the gold, and appeared not to contain any "coarse" gold. Gravity concentration may be suitable for the plant operation, but if it is, it will almost certainly require the use of a centrifugal concentrator.
- -- The tailing fractional analysis will need to be investigated in future testing, since it may be that the - 325 mesh fraction is still suffering from carry-over from a previous high grade test.

-- Previous testing, by Coastech Research, on a different sample from this property, indicated that flotation is sensitive to grind, and that overgrinding will increase the grade of the flotation tailing. This will need to be investigated further.

### RECOMMENDATIONS:

-- Prior to undertaking any additional laboratory testing, the mineral character of the "graphite" concentrate needs to be investigated and understood.

Because of the uncertain mineral composition of this product and the possible effect which it may have on the future of this property, this report has been issued in DRAFT, and will remain so until this difficulty is resolved.

(5015-16)

# DRAFT



# 10000 GRAM TEST

	SAMPLE	WEIGHT	WEIGHT	GOLD ASSAY	DISTRIBUTION	CUMULATIVE	CUMULATIVE	CUMULATIVE	UNITS
PRODUCT	TAG NO.	GRAMS	DIST.	OZ./TON	Au	WEIGHT	Au RECOVERY	Au GRADE	
								OZ./TON	
CONCENTRATE 1	67005	180.7	1.87%	0.187	13.94%	1.87%	13.94%	0.187	33.79
CONCENTRATE 2	67006	255.0	2.63%	0.149	15.68%	4.50%	29.62%	0.165	38.00
CONCENTRATE 3	67007	262.8	2.71%	0.081	8.78%	7.21%	38.40%	0.133	21.29
CONCENTRATE 4	67008	132.3	1.37%	0.058	3.17%	8.58%	41.57%	0.121	7.67
FINAL TAILING	67009	8852.0	91.42%	0.016	58.43%				141.63
OVERSIZE (+20 ME	SH)	0.0	0.00%	0.000	0.00%				0.00
CALCULATED HEAD	)	9682.8	100.00%	0.025	100.00%				242.38
ASSAYED HEAD	67004	317.2		0.035					

### 250 GRAM TEST

# 750 GRAM TEST

	SAMPLE	WEIGHT C	GOLD ASSAY			SAMPLE	WEIGHT	GOLD ASSAY
	TAG NO.	GRAMS	OZ./TON			TAG NO.	GRAMS	OZ./TON
TAIL (LIGHTS)	69073	39.8		TAIL (	(LIGHTS)	67002	121.9	
CONC. (HEAVIES)	69074	209.3	0.054	CONC.	(HEAVIES)	67003	624.1	0.031
	TOTAL	249.1			•	TOTAL	746.0	

### **500 GRAM TEST**

		SAMPLE	WEIGHT	GOLD ASSAY
		TAG NO.	GRAMS	OZ./TON
TAIL	(LIGHTS)	69075	84.2	2
CONC.	(HEAVIES)	67001	413.8	0.033
		TOTAL	498.0	)

PROCESS RI	ESEARCH ASSOCIATES LTD.	FACSIMIL			
9145 Shaughness Vancouver, B.C. Canada V6P 6R9	y Street	TEL. (604) 322-0118 FAX. (604) 322-4907			
PROJECT NO: NO. OF PAGES	93-007 (including this page): 3	DATE: March 10, 1993			
COMPANY: ATTENTION:	Quinto Mining Mr. Paul Schiller				

2 604 322 4907

BION RESEARCH

P.01

Dear Mr. Schiller:

FAX NO: FROM:

### Re: Flotation test to recover Sericite.

662-3161

Bern Klein

03/12/93 14:15

Attached are the flowsheet and mass balance for the flotation test that was performed. As was apparent, the test showed that it was difficult to float a sericite product. The poor selectivity is mainly due to the inability to reduce the pulp pH to the 3.0 - 3.5 range. At higher pH levels, the flotation collector (Armac T) activates almost all oxide minerals present. The result was that everything tried to float which produced the excessively stable froth. The difficulties associated with de-sliming also contributed to the poor selectivity. Excessive amounts of fine particles contribute to the stability of the froth which can in turn entrain gangue particles. It is also possible that the presence of the fines contributed to the difficulties associated with decreasing the pulp pH.

Based on the results of the first test, a second test is proposed. The float test should be carried out on coarser feed material to minimize the amount of slimes present. Following graphite flotation, xanthate will be added to produce a pyrite concentrate. Prior to sericite flotation, the pulp will be diluted and de-slimed. It may be necessary to deslime twice to remove a majority of the fines. For the sericite flotation the pH will be lowered to 3.5 using  $H_2SO_4$ . It is hoped that the pH can be lowered once the slimes are removed. The pulp will conditioned at a high pulp density (following desliming) using Armac T.

The products from float test one are dried and can be picked up. If you prefer, we can send the samples to an analytical laboratory for you. If you have any questions regarding the test one results or the propose float tests, please call me.

Sincerely,

Sentral Mein

Bernhard Klein, Ph.D. Senior Process Metallurgist PROCESS RESEARCH ASSOCIATES LTD.

cc. D. Drummond

03/12/93 14:16 🛛 🛣	604	322	4907
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### TESTWORK PROCEDURE

Test No: 93-007 F1

Date: 8-Mar-93

Purpose: Initial bench flotation scoping test

STAGE	TIME	ADDITIONS			
	(Minutes)	g/tonne	REAGENT		
Grínd	0				
Graphite Flotation			Natural pH ≂ 8.0		
Graphite rougher float	10	50	MIBC		
Graphite scavenger float	5	18	MIBC		
Sericite Flotation					
De-Slime			Unable to settle the fines		
Condition	15	26315	93% Conc. H2SO4 Unable to lower pH 8.0 to pH3.0		
		750	Armac T		
Rougher float	3	;	pH=4.8		
Condition	5	1500	Armac T		
Scavenger float	5		pH=6.0		
			<u> </u>		

BION RESEARCH

### MATERIAL BALANCE

Date: March 8, 1993

Project no :	93-007
Test no :	F1

90% -200 mesh material Sample description :

Products	Weight	L (%)	Assays Graphite Sericite (%) (%)	Graphite	% Distribution Sericite
Graphite Ro Conc Graphite Scav Conc Total Graphite Conc Sericite Ro Conc Sericite Scav Conc Total Sericite Conc Final Tails	106.7 55.6 1 <b>62.3</b> 68.4 549.3 <b>617.7</b> 1273.3	5.2 2.7 <b>7.9</b> 3.3 26.8 <b>30.1</b> 62.0			
Calculated head Assay head	2053.3	100.0			

# APPENDIX D

XRD, electron microprobe and optical studies By Dr. Maria Mastalerz
# THE UNIVERSITY OF BRITISH COLUMBIA



Department of Geological Sciences 6339 Stores Road Vancouver, B.C. Canada V6T 1Z4 Tel: (604) 822-2449 Fax: (604) 822-6088

Paul Schiller Quinto Mining Corporation 606-626 West Pender Street Vancouver, B.C. V6B 1V9

10th March, 1993

Dear Mr. Schiller,

As you requested I am providing the analyses of two samples containing graphite. The analyses include: XRD, electron microprobe, optical observation in transmitted and reflected light (plus reflectance measurements in reflected light) as well as dissolution in HCl and HF and subdivisions into grain size fractions. The report enclosed includes analyses and short interpretation of the analyses.

The total cost of analyses is 800 dollars and it includes :

17 hours of sample preparation (making polished section, coating samples, preparation for XRD) as well as optical studies, dissolution and splitting into size groups -  $17 \times \$30 = \$510$ , 3 hours of XRD plus microprobe (analyst time plus machine time)  $3 \times \$80 = \$240$ , materials used during sample preparation (chemicals, resin, polishing powder etc) \$ 50

Total \$800

Please make cheque payable to Geotrans Consulting Services.

If you need any further assistance with the interpretations or some other analyses, please contact me.

Yours sincerely,

Maria Massaler.

Dr. Maria Mastalerz

#### Graphite concentrate from the sandstone

METHODS:

- 1) Optical examination transmitted and reflected light
- 2) XRD
- 3) Dissolution with HCl and HF

## RESULTS

Optical studies revealed the following minerals associated with graphite: carbonates (calcite and dolomite), quartz, native sulphur and very rare green translucent mineral, possibly pyroxene.

Treatment with HCl resulted in weight loss of 3.47%. Carbonate content then is 3.47%, or a bit less than that because it is possible that some graphite was lost after treatment with HCl, during transferal from one container to the other.

Microscopic examination after HCl treatment showed that no carbonatesremainded and quartz was the dominating mineral. The portion left after HCl dissolution was separated into the following size fractions:

1) > 50  mesh	- 8.55%
2) 50-60 mesh	- 7.50%
3) 60-150 mesh	- 45.06%
4) <150 mesh	<u>- 38,89%</u>
	100.00%

The composition of each fraction is as follows:

Fraction 1. ( >50 mesh )-- flakes of graphite + 0.5% quartz + traces of sulphur

Fraction 2. ( 50-60 mesh ) - as above

Fraction 3.(60-150 mesh) - flakes of graphite +15% quartz + 0.3% sulphur + traces of

green mineral

Fraction 4. ( <150 mesh) - flakes of graphite +  $\sim 10\%$  quartz + traces of sulphur

Dissolution in HF removed all inorganic matter 0.4680g - weight before treatment 0.4020g - weight after treatment 0.0660g - difference = 14.1% weight loss during treatment with HF This 14.1% is mainly quartz below 60 mesh in size

**Reflected light microscopy and XRD** were used to determine the level of graphitization.1. Reflectance values (Romax) ranging from 16.6 to 19.8% (in oil) indicate that the material is natural graphite. These values compare well with reflectances of other natural graphite (Kwiecinska et al., 1977). The X-ray diffraction of the concentrate after treatment with HCl shows that there is exellent agreement between the theoretical peak position for graphite (002 diffraction; the green line at 20 = 26.6 degrees) and the most intense peak of the sample. Thus, the sample contains crystalline graphite. Unfortunately, it was not possible to identify other minerals with XRD because their concentrations were too low to yield discernible peak.

#### Mylonitized quartz-muscovite-calcite-graphite schist

#### METHODS:

1) optical studies (transmitted and reflected light)

2) electron microprobe

### RESULTS

Observation in reflected light showed nothing; this rock is very hard to polish. Only dispersed bright spots (i.e., areas of high reflectance) were visible but they were very small and beyond resolution at 500x magnification. Thin section analysis showed the presence of carbonaceous material, muscovite, quartz, calcite and pyrite. There are also traces of light green pyroxene in some pressure shadows. The estimated volume is as follows:

carbonaceous material (graphite)	35%
calcite	35%
muscovite	15%
quartz	10%
pyrite	3-5%

Microprobe analyses of a polished blocks of the raw sample (4 analyses, in weight %) are presented below.

An.	1	2	3	4
c	17.00	10.78	10.99	15.00
0	51.26	52.85	52.34	53,22
S	0.00	0.04	0.03	0.01
Fe	5.70	0.07	0.01	4.30
Si	22.67	25.75	27.54	18.67
AJ	2.00	8.39	7.07	1.61
Ca	1.30	2.01	1.97	5.00
Total	99.93	99.89	99.95	97.81

Microprobe analyses were done using an electron beam size of 10 micrometers. However, the elemental data (e.g., O, Si, Al contents) indicate that the beam interacted with graphite plus other minerals present in the sample. Thus, carbon data reflect contribution from graphite plus carbonates, and suggest that graphite content is below 10 weight %. Due to the complex "intergrowth" of the calcite, muscovite, quartz and graphite, it was not possible to obtain good representative data on mica composition in the sample.

Due to huge surface imperfections (hard to polish), It was possible to obtain only one microprobe analysis on the pellet from the ground concentrate from the graphite shist.

 $\mathbf{r}$ 

This analysis is as follows:

	C - 40.0070
	<b>O - 30.30%</b>
	S - 0.06%
	Fe - 0.60%
	Si - 9.91%
	Al - 4.96%
	Ca - 1.50%
Total	96.69%

18 860/

As expected, this analysis shows much higher content of carbonaceous material than the raw sample, and suggesting that graphite content in the concentrate may reach 45%.



TIMBERED ⊙ D-3 D-4 · 0 D-5 . • • •

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⊙ Survey Spad

0 5 10

